

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double arrow pointing towards the right. The substance(s) on the left side of the arrow is/are
 - reactant(s)**
 - product(s)
 - solvent
- An ionic chemical equation is represented by adding the
 - charges**
 - charges of the particles
 - number of particles
- An ionic equation shows a reaction by adding particles that are in the
 - state
 - aqueous or water**
 - solid
 - precipitate
- The activity for precipitation reactions are used to predict whether
 - soluble ionic compounds will react to form an insoluble precipitate
 - insoluble ionic compounds will react to form a soluble precipitate
 - soluble ionic compounds will react
 - insoluble ionic compounds will react to form a soluble precipitate**
- The coefficients are chemical equations
 - indicate the number of moles of each substance that react**
 - show the number of grams of each substance that react
 - show the number of moles of the substances
 - show the relative proportions of each atom
- If you're to balance the chemical equation with one variable, then
 - X will replace one of the coefficients**
 - X will replace one of the subscripts**
 - X will replace every coefficient
 - X will replace every atom

Chapter 8 Test B

Robert R. Redfield



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