

## Chapter 9: Chemical Reactions

- \_\_\_\_\_ 1. Which of the following could be an indication of a chemical reaction of a compound or element?
- It is a solid at room temperature.
  - It melts at 631<sup>o</sup>C.
  - It expands upon freezing.
  - It burns in an atmosphere of chlorine.
- \_\_\_\_\_ 2. In this unbalanced chemical reaction, which components are the reactants?
- $$C_7H_{16} + O_2 \rightarrow CO_2 + H_2O$$
- C<sub>7</sub>H<sub>16</sub> and O<sub>2</sub>
  - C<sub>7</sub>H<sub>16</sub> and CO<sub>2</sub>
  - C<sub>7</sub>H<sub>16</sub> and H<sub>2</sub>O
  - CO<sub>2</sub> and H<sub>2</sub>O
- \_\_\_\_\_ 3. When Ca(OH)<sub>2</sub> reacts with H<sub>2</sub>SO<sub>4</sub>, what are the products.
- Ca(OH)<sub>2</sub> and H<sub>2</sub>SO<sub>4</sub>
  - CaH<sub>2</sub> and H<sub>2</sub>SO<sub>4</sub>
  - CaSO<sub>4</sub> and H<sub>2</sub>O
  - CaO and H<sub>2</sub>SO<sub>4</sub>
- Hint: This is a double-replacement reaction, positive ions switch places.
- \_\_\_\_\_ 4. What are the products of the reaction between barium hydroxide and hydrochloric acid?
- Barium chloride and water
  - Barium hydroxide and water
  - Barium chloride and barium hydride
  - Chloric acid and barium hydroxide
- Hint: Write out the skeleton equation. This is a double-replacement reaction.
- \_\_\_\_\_ 5. This symbol separates the reactants from products and means "to yield".
- +
  - (s)
  - 
  - (g)
- \_\_\_\_\_ 6. In a chemical equation, the number written in front of the chemical formula is the
- Reactant
  - Subscript
  - Equation
  - Coefficient
- \_\_\_\_\_ 7. When there are exactly the same number and type of atoms on both sides of a chemical equation, the equation is \_\_\_\_\_.
- Balanced
  - Ionic
  - Synthetic
  - Decomposed
- \_\_\_\_\_ 8. The chemical equation  $CaCO_3 \rightarrow CaO + CO_2$  is classified as a
- Synthesis reaction
  - Decomposition reaction
  - Single-replacement reaction
  - Double-replacement reaction
- \_\_\_\_\_ 9. In the combustion reaction of propane (C<sub>3</sub>H<sub>8</sub>) with oxygen gas, what are the products?
- C<sub>3</sub>H<sub>8</sub> and O<sub>2</sub>
  - C<sub>3</sub>H<sub>8</sub> and CO<sub>2</sub>
  - O<sub>2</sub> and H<sub>2</sub>O
  - H<sub>2</sub>O and CO<sub>2</sub>
- \_\_\_\_\_ 10. The major physical indicator(s) of a combustion reaction is

# Chapter 9 Chemical Reactions Test

**Marco Casella**



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