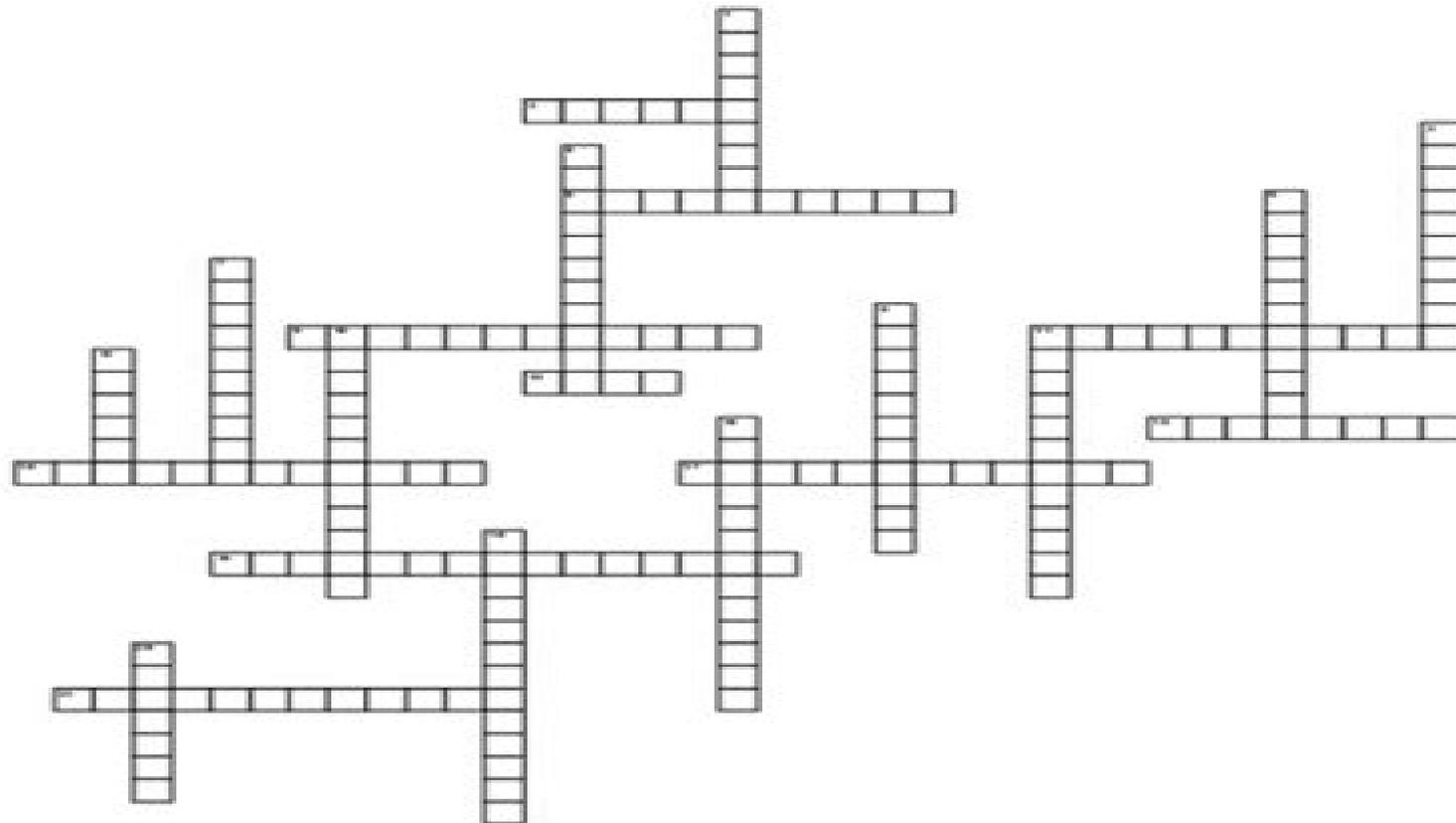


Thermal Energy



Across

- [2] ionized gas that is a combination of the negatively charged free electrons and positively charged ions
 [5] heat related to the phase change of a substance rather than a change of temperature
 [6] phase change from liquid to gas
 [11] phase change from solid to gas
 [13] transfer of thermal (or internal) energy due to a temperature difference
 [15] phase change from liquid to solid
 [16] phase change from gas to liquid
 [17] amount of heat necessary to change the temperature of a substance by 1.00 °C
 [19] temperature scale in which the freezing point of water is 32
 [21] temperature scale in which the freezing point of water is 0 °C

Down

- [1] energy transferred by electromagnetic waves
 [3] heat transfer by the movement of fluid
 [4] temperature scale in which 0 K is the lowest possible temperature, representing absolute zero
 [8] quantity measured by a thermometer
 [7] heat transfer through stationary matter by physical contact
 [9] transition between solid, liquid, or gas states of a substance
 [10] lowest possible temperature; the temperature at which all molecular motion ceases
 [11] amount of heat necessary to change the temperature of 1.00 kg of a substance by 1.00 °C
 [12] unit on the Kelvin temperature scale; note that it is never referred to in terms of "degrees" Kelvin
 [14] average random kinetic energy of a molecule or an atom
 [18] unit on the Celsius temperature scale
 [20] phase change from solid to liquid

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