

Thermodynamic Equilibrium - 1 (Application of Equilibrium)

Thermodynamic Equilibrium

Thermodynamic Equilibrium: A system is in thermodynamic equilibrium if its macroscopic properties do not change with time.

1. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the temperature is constant.
2. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the pressure is constant.
3. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the volume is constant.
4. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the number of particles is constant.
5. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the internal energy is constant.
6. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the entropy is constant.
7. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the Gibbs free energy is constant.
8. In a closed system at constant volume and constant pressure, the system is in thermodynamic equilibrium if the Helmholtz free energy is constant.

Chapter Test Thermal Energy Answers 16

SATYAM SIR



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