

Chapter Test II

A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

Column I

1. An oxidation-reduction reaction in which an element is oxidized and another element is reduced.
2. An oxidation-reduction reaction in which a substance is oxidized and the same substance is reduced.
3. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
4. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
5. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
6. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.

Column II

- a. oxidation-reduction
- b. redox reaction
- c. disproportionation
- d. combustion
- e. synthesis
- f. decomposition

B. Multiple Choice

Choose the best answer for each question.

1. Which of the following is a redox reaction?
 - a. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - b. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - c. $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$
 - d. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
2. Which of the following is a redox reaction?
 - a. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - b. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - c. $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$
 - d. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
3. Which of the following is a redox reaction?
 - a. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - b. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - c. $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$
 - d. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

Chemistry Chapter 12 Stoichiometry Test

Albert A Gayle



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