

Chapter 11: Changes in Matter ---Study Guide

These items can be found in your child's packet in the science section or in their science book. All items have been discussed at length in class. Please refer to the cover of the packet to view which packet pages to study.

Words to Know:

physical change chemical change states of matter mixture solution

*Here is a link to help your child to study the vocabulary for Chapter 11.

<http://quizlet.com/1022509/scott-foresman-science-grade-3-chapter-11-flash-cards/>

(This link and other links are posted on your science teacher's website.)

Ideas to Know:

- **The states of matter** are solids, liquids, and gases.
- **Physical change** is when the way matter looks is changed, but it does not become a new kind of matter.
- **Some examples of physical changes include:**
cutting or folding paper, making something out of Play-doh, cutting up food, folding clothes
- **Chemical change** is when one kind of matter changes into a different kind of matter.
- **Some examples of chemical changes include:**
baking bread, rust forming, burning paper or wood, soaps used to clean, batteries releasing electricity, burning gasoline in cars
- **Matter can change from one state to another** (from a solid to a liquid to a gas), but this is NOT a chemical change. For example, ice melting to a liquid is a physical change because **NO** new matter is formed.
- A **mixture** is made of 2 or more kinds of matter that are placed together. Example: tossed salad, fruit salad
- **The types of matter in a mixture do not change into another substance**. Each kind of matter can be separated from the other kinds of matter in the mix.
- A **solution** is when a substance dissolves into another substance.
- A saltwater mixture can be separated by boiling the water. The boiling water evaporates into the air to become water vapor (a gas), and the salt is left behind.

Chemistry Matter And Change Chapter 11 Assessment Answers

Christian Drosten



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