

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O₂ 1.

How many moles of oxygen are produced from twelve moles of Potassium chlorate?

21 mol O₂ 2.

How many moles of oxygen are produced from 14 moles of Potassium chlorate?

45 mol H₂O 3.

How many moles of water are produced from 22.5 moles of oxygen?

20 mol HCl 4.

Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

21 mol Cl₂ 5.

Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



Chemistry Section 1 Review Stoichiometry Answers

Lingsheng Yao



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A Complete Preparation for the MCAT Aftab S. Hassan, James L. Flowers, 1992 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities There are details on setting up a self managed study programme with guidelines for time management and stress management All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress

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