

**Atomic Structure & Theory**

1. Can the number of protons in an element ever change?

- A. No, protons and neutrons are bound together in atomic nuclei, and neither can change.
- B. Yes, isotopes of the same element have different numbers of protons.
- C. No, if the number of protons changes, the element changes.
- D. Yes, when different elements bond, they can share or transfer protons.
- 

**Atomic Structure & Theory**

2. All matter, both living and nonliving, is made up of tiny particles called \_\_\_\_\_.

- A. cells
- B. atoms
- C. dust
- D. bacteria
- 

**The Periodic Table**

4. Which of the following is true of all the elements in group 2 on the periodic table?

- I. They are all metals.
- II. They all have two electrons in their outermost shell.
- III. They all tend to lose electrons and become positive ions.
- IV. They all have similar chemical properties.
- A. II, III, and IV only
- B. I, III, and IV only
- C. I, II, III, and IV
- D. III and IV only
-

# Chemistry Semester 1 Exam Review Study Island

**Shasha Hu**



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