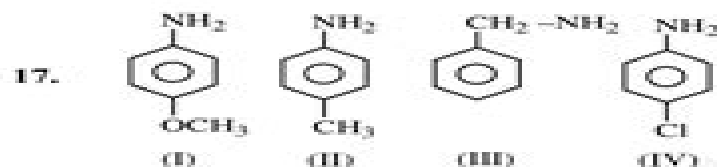


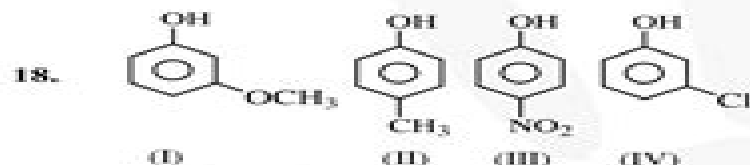
16. Which of the following compounds is most basic ? [AIPMT Mains -2011]



The correct decreasing order of pK_b is :-

- (1) I > II > III > IV (2) III > IV > II > I
 (3) II > III > IV > I (4) IV > II > I > III

[AIIMS-2011]



The correct decreasing order of pK_a is :-

- (1) II > I > IV > III (2) IV > II > III > I
 (3) III > II > IV > I (4) IV > I > II > III

[AIIMS-2011]

19. The correct order of decreasing acid strength of trichloroacetic acid (A), trifluoroacetic acid (B), acetic acid (C) and formic acid (D) is: [AIPMT Pre.-2012]

- (1) A > B > C > D (2) A > C > B > D
 (3) B > A > D > C (4) B > D > C > A

20. pK_a increases in benzoic acid, substituent "x" bonded at para position then "x" is -

- (1) $-\text{COOH}$ (2) $-\text{NO}_2$
 (3) $-\text{CN}$ (4) $-\text{OCH}_3$

[AIIMS 2012]

21. The radical,  $\text{CH}_2\cdot$ is aromatic because it has :- [NEET-UG 2013]

- (1) 6p-orbitals and 7 unpaired electrons
 (2) 6p-orbitals and 6 unpaired electrons
 (3) 7p-orbitals and 6 unpaired electrons
 (4) 7p-orbitals and 7 unpaired electrons

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Exploring Chemistry (Loose-Leaf) Matthew Johll, 2012-05-04 Matthew Johll's Exploring Chemistry covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It's the same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book's specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more

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Government Reports Announcements & Index, 1993

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to think about organic chemistry in new and exciting ways Based on the author's first hand classroom experience the text uses complementary conceptual models to give new perspectives on the structures and reactions of organic compounds The first five chapters of the text discuss the structure and bonding of stable molecules and reactive intermediates These are followed by a chapter exploring the methods that organic chemists use to study reaction mechanisms The remaining chapters examine different types of acid base substitution addition elimination pericyclic and photochemical reactions This Second Edition has been thoroughly updated and revised to reflect the latest findings in physical organic chemistry Moreover this edition features New references to the latest primary and review literature More study questions to help readers better understand and apply new concepts in organic chemistry Coverage of new topics including density functional theory quantum theory of atoms in molecules Marcus theory molecular simulations effect of solvent on organic reactions asymmetric induction in nucleophilic additions to carbonyl compounds and dynamic effects on reaction pathways The nearly 400 problems in the text do more than allow students to test their understanding of the concepts presented in each chapter They also encourage readers to actively review and evaluate the chemical literature and to develop and defend their own ideas With its emphasis on complementary models and independent problem solving this text is ideal for upper level undergraduate and graduate courses in organic chemistry

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