



COVALENT BONDING

Practice Problems

Do your answers to the following problems.

SECTION 8.1 MOLECULAR COMPOUNDS

- Draw the Lewis structure for each of the following molecules.
 - H_2O
 - CO_2
 - H_2S
 - H_2SO_4
 - H_2SO_3
 - H_2SO_2
- Draw the Lewis structure for each of the following molecules.
 - H_2O_2
 - H_2O_3
 - H_2O_4
- Draw the Lewis structure for each of the following molecules.
 - H_2O_2
 - H_2O_3
 - H_2O_4
- Draw the Lewis structure for each of the following molecules.
 - H_2O_2
 - H_2O_3
 - H_2O_4
- Draw the Lewis structure for each of the following molecules.
 - H_2O_2
 - H_2O_3
 - H_2O_4
- Draw the Lewis structure for each of the following molecules.
 - H_2O_2
 - H_2O_3
 - H_2O_4

SECTION 8.2 THE NATURE OF COVALENT BONDING

- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{F}$
- Draw the Lewis structure for each of the following molecules.

$\begin{array}{c} \text{H} \\ | \\ \text{H}-\text{C}-\text{H} \\ | \\ \text{H} \end{array}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$



SECTION 8.3 BONDING: THE ORBITALS

- Draw the Lewis structure for each of the following molecules.
 - H_2
 - O_2
 - N_2
 - CO
 - NO
 - NO_2
 - NO_3
 - O_3
 - SO_2
 - SO_3
 - SO_4^{2-}
 - PO_4^{3-}
 - PO_3^{3-}
 - PO_2^{3-}
 - PO^{3-}
 - SiO_4^{4-}
 - SiO_3^{4-}
 - SiO_2^{4-}
 - SiO^{4-}
 - SiO_2
 - SiO
 - SiO_3
 - SiO_4
 - SiO_5
 - SiO_6
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 - SiO_{100}
- Draw the Lewis structure for each of the following molecules.
 - H_2
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 - CO
 - NO
 - NO_2
 - NO_3
 - O_3
 - SO_2
 - SO_3
 - SO_4^{2-}
 - PO_4^{3-}
 - PO_3^{3-}
 - PO_2^{3-}
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