

### Chemistry 12 Homework Assignment 3-1

46

Name

KEY

Due Date

Correct and Hand In By

This worksheet covers material from Tutorials 7 and 8, class notes and SW pages 73-87.

These questions will be marked and the mark recorded in the "Homework" section in the mark book. One set of corrections will be allowed. Get help if you need it.

1. Decide whether each of the following are electrolytes or not (yes or no) when dissolved in water. (5 marks)

a)  $\text{AgNO}_3$  ..... yesb)  $\text{CH}_3\text{CH}_2\text{OH}$  ..... noc)  $\text{NH}_4\text{Cl}$  ..... yesd)  $\text{CH}_3\text{CH}_2\text{COOH}$  ..... yese)  $\text{Ba}(\text{OH})_2$  ..... yes

2. Give three examples of ionic compounds (formulas must be correct) (3 marks)

1. \_\_\_\_\_ 2. \_\_\_\_\_ 3. \_\_\_\_\_

3. Give two examples of molecular compounds (formulas must be correct) (2 marks)

1. \_\_\_\_\_ 2. \_\_\_\_\_

Define a saturated solution (1 mark)

Solubility is the equilibrium concentration of a substance in solution at a given temperature. (1 mark)

6. Write a net ionic equation (balanced, all charges, subscripts, arrow(s) correct, correct order etc.) for dissolving ammonium sulphate. (2 marks)



# Chemistry Unit 3 Test Answer Key

**Anthony S. Fauci**



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depth of

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web summary of test method 3 1 this test method permits hardness measurements based on either initial indentation or

indentation after a specified period of time or both durometers with maximum reading indicators used to determine maximum hardness values of a material may yield lower hardness when the maximum indicator is used

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