

CHAPTER 1 PRELIMINARIES

1.1 REAL NUMBERS AND THE REAL LINE

1. Executing long division, $\frac{1}{3} = 0.\overline{3}$, $\frac{2}{3} = 0.\overline{6}$, $\frac{1}{4} = 0.25$, $\frac{3}{4} = 0.75$, $\frac{1}{5} = 0.2$
2. Executing long division, $\frac{1}{11} = 0.\overline{09}$, $\frac{2}{11} = 0.\overline{18}$, $\frac{3}{11} = 0.\overline{27}$, $\frac{4}{11} = 0.\overline{36}$, $\frac{5}{11} = 0.\overline{45}$
3. NT = necessarily true, NNT = Not necessarily true. Given: $2 < x < 6$.
 - a) NNT. 5 is a counter example.
 - b) NT. $2 < x < 6 \Rightarrow 2 - 2 < x - 2 < 6 - 2 \Rightarrow 0 < x - 2 < 4$.
 - c) NT. $2 < x < 6 \Rightarrow 2/2 < x/2 < 6/2 \Rightarrow 1 < x < 3$.
 - d) NT. $2 < x < 6 \Rightarrow 1/2 > 1/x > 1/6 \Rightarrow 1/6 < 1/x < 1/2$.
 - e) NT. $2 < x < 6 \Rightarrow 1/2 > 1/x > 1/6 \Rightarrow 1/6 < 1/x < 1/2 \Rightarrow 6(1/6) < 6(1/x) < 6(1/2) \Rightarrow 1 < 6/x < 3$.
 - f) NT. $2 < x < 6 \Rightarrow x < 6 \Rightarrow (x - 4) < 2$ and $2 < x < 6 \Rightarrow x > 2 \Rightarrow -x < -2 \Rightarrow -x + 4 < 2 \Rightarrow -(x - 4) < 2$.
The pair of inequalities $(x - 4) < 2$ and $-(x - 4) < 2 \Rightarrow |x - 4| < 2$.
 - g) NT. $2 < x < 6 \Rightarrow -2 > -x > -6 \Rightarrow -6 < -x < -2$. But $-2 < 2$. So $-6 < -x < -2 < 2$ or $-6 < -x < 2$.
 - h) NT. $2 < x < 6 \Rightarrow -1(2) > -1(x) < -1(6) \Rightarrow -6 < -x < -2$

4. NT = necessarily true, NNT = Not necessarily true. Given: $-1 < y - 5 < 1$.
 - a) NT. $-1 < y - 5 < 1 \Rightarrow -1 + 5 < y - 5 + 5 < 1 + 5 \Rightarrow 4 < y < 6$.
 - b) NNT. $y = 5$ is a counter example. (Actually, never true given that $4 < y < 6$)
 - c) NT. From a), $-1 < y - 5 < 1 \Rightarrow 4 < y < 6 \Rightarrow y > 4$.
 - d) NT. From a), $-1 < y - 5 < 1 \Rightarrow 4 < y < 6 \Rightarrow y < 6$.
 - e) NT. $-1 < y - 5 < 1 \Rightarrow -1 + 1 < y - 5 + 1 < 1 + 1 \Rightarrow 0 < y - 4 < 2$.
 - f) NT. $-1 < y - 5 < 1 \Rightarrow (1/2)(-1 + 5) < (1/2)(y - 5 + 5) < (1/2)(1 + 5) \Rightarrow 2 < y/2 < 3$.
 - g) NT. From a), $4 < y < 6 \Rightarrow 1/4 > 1/y > 1/6 \Rightarrow 1/6 < 1/y < 1/4$.
 - h) NT. $-1 < y - 5 < 1 \Rightarrow y - 5 > -1 \Rightarrow y > 4 \Rightarrow -y < -4 \Rightarrow -y + 5 < 1 \Rightarrow -(y - 5) < 1$.
Also, $-1 < y - 5 < 1 \Rightarrow y - 5 < 1$. The pair of inequalities $-(y - 5) < 1$ and $(y - 5) < 1 \Rightarrow |y - 5| < 1$.

5. $-2x > 4 \Rightarrow x < -2$



6. $8 - 3x \geq 5 \Rightarrow -3x \geq -3 \Rightarrow x \leq 1$



7. $5x - 3 \leq 7 - 3x \Rightarrow 8x \leq 10 \Rightarrow x \leq \frac{5}{4}$



8. $3(2 - x) > 2(3 + x) \Rightarrow 6 - 3x > 6 + 2x$
 $\Rightarrow 0 > 5x \Rightarrow 0 > x$



9. $2x - \frac{1}{4} \geq 7x + \frac{1}{4} \Rightarrow -\frac{1}{4} - \frac{1}{4} \geq 5x$
 $\Rightarrow \frac{1}{2} \left(-\frac{2}{4}\right) \geq x$ or $-\frac{1}{4} \geq x$



10. $\frac{3x}{4} < \frac{2x-4}{2} \Rightarrow 12 - 2x < 12x - 16$
 $\Rightarrow 28 < 14x \Rightarrow 2 < x$



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H Kauffman



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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton $ha + b \rightarrow hb$ strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the K_a or K_b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web spreadsheet computations prelab report this is the type of computations the computer does to determine first and second derivatives refer to these spreadsheet columns for the comments below ph titration graph lab data provide columns a

and b

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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through the use of a primary standard in this experiment we will be using naoh and hcl as well as khp

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