

Name _____ Date _____ Class _____

CHAPTER 13 STUDY GUIDE FOR CONTENT MASTERY

States of Matter

Section 13.1 Gases

In your textbook, read about the kinetic molecular theory.

Complete each statement.

- The kinetic molecular theory describes the behavior of gases in terms of particles in motion.
- The kinetic molecular theory makes the following assumptions:
 - In a sample of a gas, the volume of the gas particles themselves is very small compared to the volume of the sample.
 - Between gas particles and the walls, there are no significant attractive or repulsive forces between gas particles.
 - Gas particles are in constant and random motion.
 - The collisions between gas particles are elastic, that is, no kinetic energy is lost.
- The kinetic energy of a particle is represented by the equation $KE = \frac{1}{2}mv^2$.
- Temperature is a measure of the average kinetic energy of the particles in a sample of matter.

In your textbook, read about explaining the behavior of gases.

For each statement below, write true or false.

- | | |
|--------------|--|
| <u>True</u> | 1. Gases are less dense than solids because there is a lot of space between the particles of a gas. |
| <u>True</u> | 2. The random motion of gas particles causes a gas to expand until it fills its container. |
| <u>False</u> | 3. The density of a gas decreases as it is compressed. |
| <u>True</u> | 4. A gas can flow into a space occupied by another gas. |
| <u>True</u> | 5. The diffusion of a gas is caused by the random motion of the particles of the gas. |
| <u>False</u> | 6. Lighter gas particles diffuse less rapidly than do heavier gas particles. |
| <u>True</u> | 7. During effusion, a gas escapes through a tiny opening into a vacuum. |
| <u>False</u> | 8. Graham's law of effusion states that the rate of effusion for a gas is directly related to the square root of its molar mass. |

Study Guide for Content Mastery

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CHAPTER 13 STUDY GUIDE FOR CONTENT MASTERY

Section 13.1 continued

In your textbook, read about gas pressure.

Circle the letter of the choice that best completes the statement or answers the question.

- Pressure is defined as force per unit
 - area
 - time
 - mass
 - volume
- What is an instrument designed to measure atmospheric pressure?
 - barometer
 - manometer
 - physoconimeter
 - densimeter
- The height of the liquid in a barometer is affected by all of the following EXCEPT the
 - altitude.
 - density of the liquid in the column.
 - atmospheric pressure.
 - diameter of the column tube.
- The pressure of the gas in a manometer is directly related to which of the following quantities?
 - height of the mercury column in the closed-end arm
 - height of the mercury column in the open-end arm
 - $h + 760$
 - $h - 760$
- One atmosphere is equal to a pressure of
 - 76 mm Hg
 - 101.3 kPa
 - 147 psi
 - 760 mm
- The partial pressure of a gas depends on all of the following EXCEPT the
 - concentration of the gas.
 - size of the container.
 - density of the gas.
 - temperature of the gas.
- The pressure of a sample of air in a manometer is 101.3 kPa. What is the partial pressure of nitrogen (N_2) in the sample if the combined partial pressures of the other gases is 22.4 kPa?
 - 42.4 kPa
 - 78.9 kPa
 - 101.3 kPa
 - 123.7 kPa

Use the figure to answer the following questions.

- What instrument is illustrated in the figure? barometer
- Who invented this instrument? Evangelista Torricelli
- What are the two opposing forces that control the height of the mercury in the column? gravity and atmospheric pressure
- What does it mean when the level of mercury drops in the column? The atmospheric pressure has increased.



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on ne badine pas avec l amour dossier pédagogique de justine francioli mis à jour par marc stéphan pour rendre camille jalouse perdican s amuse à séduire rosette la soeur de lait de celle qu il aime lorsque camille découvre le stratagème de son amoureux elle décide d entrer à son tour dans ce jeu dangereux

on ne badine pas avec l amour musset analyse complète - Jul 06 2022

publiée pour la première fois en 1834 on ne badine pas avec l amour est une des œuvres les plus appréciées d alfred de musset notamment connue pour être le reflet de sa relation avec l auteure george sand pur mélange entre comédie et tragédie et entre réalité et fiction cette pièce a connu une première adaptation

musset on ne badine pas avec l amour Études littéraires - Jun 17 2023

le drame des rêves de la jeunesse conclusion le drame de l orgueil camille et perdican en effet étaient promis l un à l autre depuis longtemps et la rencontre trop habilement combinée par le baron devait préluder à un mariage heureux mais chacun d eux repousse à son tour l heureux projet en se laissant aller à un mouvement de vanité

on ne badine pas avec l amour alfred de musset bac de - Jan 12 2023

résumé de on ne badine pas avec l amour présentation de on ne badine pas avec l amour alfred de musset voici les différents commentaires de on ne badine pas avec l amour la scène d exposition acte i scène 1 le monologue de bridaine acte ii scène 2 le couple phare fin de l acte ii scène 5

résumé on ne badine pas avec l amour superprof - Aug 07 2022

celui ci arrive à ses fins mais lors de leur premier baiser rosette amoureuse de perdican les aperçoit et pousse un cri strident rosette se tue et camille mise au courant du stratagème de perdican le quitte le titre sonne comme un avertissement on

on ne badine pas avec l amour 1ère kartable - Mar 14 2023

alfred de musset 1861 la pièce se déroule au château du baron camille la nièce du baron sort du couvent elle a 18 ans elle retrouve le fils de son oncle perdican qui a 21 ans et vient d obtenir un doctorat ils ne s étaient pas vus depuis dix ans le baron souhaite les marier perdican et camille s aiment depuis toujours