

## Stoichiometry Practice Worksheet

### Balancing Equations and Simple Stoichiometry

Balance the following equations:

- 1)  $\underline{\quad} \text{N}_2 + \underline{\quad} \text{F}_2 \rightarrow \underline{\quad} \text{NF}_3$
- ✓ 2)  $\underline{\quad} \text{C}_6\text{H}_{10} + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{CO}_2 + \underline{\quad} \text{H}_2\text{O}$
- ✓ 3)  $\underline{\quad} \text{HBr} + \underline{\quad} \text{KHCO}_3 \rightarrow \underline{\quad} \text{H}_2\text{O} + \underline{\quad} \text{KBr} + \underline{\quad} \text{CO}_2$
- ✓ 4)  $\underline{\quad} \text{GaBr}_3 + \underline{\quad} \text{Na}_2\text{SO}_3 \rightarrow \underline{\quad} \text{Ga}_2(\text{SO}_3)_3 + \underline{\quad} \text{NaBr}$
- 5)  $\underline{\quad} \text{SnO} + \underline{\quad} \text{NF}_3 \rightarrow \underline{\quad} \text{SnF}_2 + \underline{\quad} \text{N}_2\text{O}_3$

Solve the following stoichiometry grams-grams problems:

- ✓ 6) Using the following equation:



How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you have an excess of sulfuric acid?

↗ Using the following equation:



How many grams of lithium nitrate will be needed to make 250 grams of lithium sulfate, assuming that you have an adequate amount of lead (IV) sulfate to do the reaction?

## 8a Stoichiometry Extra Practice Problems Answers

**C. Schlenker, Jean Dumas, Milton  
Greenblatt, Sander van Smaalen**

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*Physics and Chemistry of Low-Dimensional Inorganic Conductors* C. Schlenker, Jean Dumas, Milton Greenblatt, Sander van Smaalen, 2012-12-06 The field of low dimensional conductors has been very active for more than twenty years It has grown continuously and both the inorganic and organic materials have remarkable properties such as charge and spin density waves and superconductivity The discovery of superconductivity at high temperature in copper based quasi two dimensional conducting oxides nearly ten years ago has further enlarged the field and stimulated new research on inorganic conductors It was obviously impossible to cover such a broad field in a ten day Institute and it seemed pertinent to concentrate on inorganic conductors excluding the high T<sub>c</sub> superconducting oxides In this context it was highly desirable to include both physics and chemistry in the same Institute in order to tighten or in some cases to establish links between physicists and chemists This Advanced Study Institute is the continuation of a series of similar ones which have taken place every few years since 1974 73 participants coming from 13

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**Bite-Sized Chemistry Calculations** John Lambert, 2024-07-10 Bite Sized Chemistry Calculations is a series of books on chemistry calculations aimed at helping students overcome the challenges associated with tackling the various types of calculations encountered in different aspects of chemistry focusing on a few topics at a time to facilitate comprehension Written by an experienced chemistry educator each book in the series has been tailored to fully meet the needs of students at all levels especially those taking college level general chemistry courses as well as those following various O level curricula worldwide This part of the series explores the different types of problems and calculations encountered in mass the mole and stoichiometry including the determination of formulae of ionic compounds relative formula masses mass and percent compositions of compounds all aspects of mole calculations empirical and molecular formulae calculations based on chemical equations limiting reagents gas stoichiometry and percent yield The series is packed with many salient features that are meant to facilitate both teaching and learning Some of these include helpful explanations many examples alternative ways to solve problems plenty of practice questions complete answers an appendices With this book you will be well prepared for your exams and boost your performance

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