

Stoichiometry Worksheet 2
Chemistry

Name _____

Date _____ Hour _____

1. In the reaction : $2\text{Al}_2\text{O}_{3(s)} \rightarrow 4\text{Al}_{(s)} + 3\text{O}_{2(g)}$, the mole ratios are:

Al_2O_3 to Al _____ Al_2O_3 to O_2 _____ Al to O_2 _____ O_2 to Al _____

2. In the reaction: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the mole ratio of oxygen to water is: _____

3. In the reaction: $\text{Ca} + \text{Cl}_2 \rightarrow \text{CaCl}_2$, the mole ratio of chlorine to calcium is: _____

4. In the reaction: $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$, the mole ratio of zinc to sulfuric acid is:

5. The equation for the Haber process is represented by:



To convert 9.0 moles of hydrogen to ammonia would require how many moles of nitrogen?

6. In the equation: $2\text{KClO}_{3(g)} \rightarrow 2\text{KCl}_{(s)} + 3\text{O}_{2(g)}$, how many moles of oxygen are produced when 3.00 moles of KClO_3 decompose completely?

7. For the reaction: $\text{C}_{(s)} + 2\text{H}_{2(g)} \rightarrow \text{CH}_{4(g)}$, how many moles of H_2 are required to produce 10.0 moles of CH_4 ?

Chapter 9 Stoichiometry Section 2 Worksheet

Leo J. Malone



Chapter 9 Stoichiometry Section 2 Worksheet:

Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Chapter 9 Stoichiometry Section 2 Worksheet**

In a global inundated with monitors and the cacophony of quick communication, the profound energy and mental resonance of verbal artistry often fade into obscurity, eclipsed by the continuous barrage of noise and distractions. However, situated within the musical pages of **Chapter 9 Stoichiometry Section 2 Worksheet**, a fascinating work of fictional beauty that impulses with raw thoughts, lies an wonderful trip waiting to be embarked upon. Composed by a virtuoso wordsmith, this enchanting opus books readers on a mental odyssey, softly exposing the latent potential and profound influence embedded within the intricate internet of language. Within the heart-wrenching expanse with this evocative analysis, we can embark upon an introspective exploration of the book is main subjects, dissect their captivating publishing model, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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