

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double underline because reactants are written after the formation of a
 - product**
 - reactant molecule
 - solvent
- An ionic chemical equation is represented by adding the
 - charges**
 - charges of the particles
 - number of particles
- An ionic equation that is written as aq will describe the solution of a
 - solid
 - dissolved ionic**
 - solid
 - substance
- The activity for precipitation reactions on pages 100-101 is that
 - soluble ionic compounds are substituted for the soluble ionic compounds
 - insoluble ionic compounds
 - soluble ionic compounds
 - insoluble ionic compounds are substituted for the soluble ionic compounds**
- The coefficients for chemical equations
 - indicate the number of moles of each substance that react**
 - show the number of grams of each substance that react
 - show the number of molecules of the substances
 - show the relative proportions of each atom
- If $2H_2O(l)$ is shown above a chemical equation with one variable, then
 - 2 molecules of water will be produced**
 - 2 water molecules will be produced
 - 2 water molecules will react
 - 2 molecules of water will react

Chapter 8 Test B

**Zenos Edmund Scott,Randolph
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Glencoe Law and Ethics Medical Careers Karen Judson, Sharon Blesie-Hicks, 1998-12
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Standardized Exercises in United States History Sturgiss Brown Davis, 1921
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