



7. What is the maximum mass of aluminum hydroxide that can be produced when 25.0 mL of 0.248 M sodium hydroxide solution are mixed with 25.0 mL of 0.189 M aluminum nitrate solution?

Find mol of each reactant

$$0.0250 \text{ L} \times 0.248 \frac{\text{mol}}{\text{L}} = 6.2 \times 10^{-3} \text{ mol NaOH}$$

$$0.0250 \text{ L} \times 0.189 \frac{\text{mol}}{\text{L}} = 4.73 \times 10^{-3} \text{ mol Al}(\text{NO}_3)_3$$

$$6.20 \times 10^{-3} \text{ mol NaOH} \times \frac{1 \text{ mol Al}(\text{OH})_3}{3 \text{ mol NaOH}} \times \frac{78.0 \text{ g Al}(\text{OH})_3}{\text{mol}} =$$

$$\boxed{0.161 \text{ g Al}(\text{OH})_3}$$

need 3 times as much NaOH as Al(NO₃)₃ but we only have 1.3 x as much NaOH is limiting

8. Indicate which of the following are redox reactions. For those that are redox reactions, label the reactant that is the oxidizing agent and the reactant that is the reducing agent. You do not have to balance these reactions!



Redox agent agent
 red. agent ox. agent

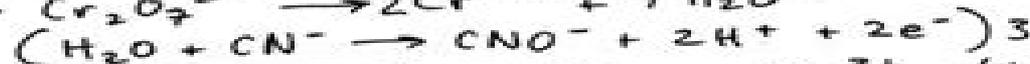


Not redox

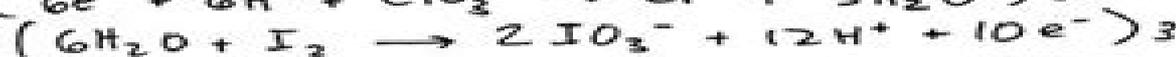


red agent ox agent

9. Aqueous dichromate ion reacts with aqueous cyanide ion (CN⁻) to produce chromium (III) ion and aqueous cyanate (CNO⁻) ion. Balance the equation using the half-reaction method, assuming the reaction occurs in acidic solution.



10. Balance the following redox reaction in basic solution.



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Frank L. Wiseman



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Children's Books in Print R R Bowker Publishing, Bowker, 1999-12

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Secondary Teaching Methods Kenneth T. Henson, 1981 *Physical Chemistry* Thomas Engel, 2006 *Proceedings of the 2024 9th International Conference on Modern Management, Education and Social Sciences (MMET 2024)* Chunyi Lin, Fong Peng Chew, Intakhab Alam Khan, Catalin Popescu, 2024-12-07 This is an open access book As a leading role in the global megatrend of scientific innovation China has been creating a more and more open environment for scientific innovation increasing the depth and breadth of academic cooperation and building a community of innovation that benefits all Such endeavors are making new contributions to the globalization and creating a community of shared future To adapt to this changing world and China's fast development in the new era 2024 9th International Conference on Modern Management Education and Social Sciences MMET 2024 to be held in September 20 22 2024 This conference takes bringing together global wisdom in scientific innovation to promote high quality development as the theme and focuses on cutting edge research fields including Modern Management and Education Technology MMET 2023 encourages the exchange of information at the forefront of research in different fields connects the most advanced academic resources in China and the world transforms research results into industrial solutions and brings together talent technology and capital to drive development The conference sincerely invites experts scholars business people and other relevant personnel from

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