



REACTING MASS CALCULATIONS - INTRODUCTION

- Step 1** Write ✓ for the substance whose mass is given and ? for the substance whose mass is to be calculated on the balanced equation
- Step 2** Find the moles of the ✓ substance (using $\text{moles} = \frac{\text{mass}}{M_r}$)
- Step 3** Use the balanced equation and your answer from step 2 to find the moles of the ? substance
- Step 4** Find the mass of the ? substance (using $\text{mass} = M_r \times \text{moles}$)

1) What mass of oxygen reacts with 12 g of magnesium?



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2) What mass of calcium hydroxide is made from 14 kg of calcium oxide?



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3) What mass of aluminium is needed to react with 640 g of iron oxide?



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4) What mass of titanium chloride reacts with 460 g of sodium?



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