



Name:

Date:

## Student Exploration: Calorimetry Lab

**Directions:** Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

**Vocabulary:** calorie, calorimeter, joule, specific heat capacity

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. The Latin word *calor* means "heat," and *meter* comes from the Greek word meaning "to measure." What do you think a **calorimeter** does?

measuring the heat of chemical reactions or physical changes as well as heat capacity

2. Where have you heard the word **calorie** before? What do you think a calorie is?

calorie, a unit of energy or heat variously defined. the calorie was originally defined as the amount of heat required at a pressure of 1 standard atmosphere to raise the temperature of 1 gram of water 1 degree celsius.

### Gizmo Warm-up

A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Calorimeters can be used to find a substance's **specific heat capacity**. You will use the *Calorimetry Lab* Gizmo to determine the specific heat capacities of various substances.



1. On the **SIMULATION** pane, select **Copper**. Use the slider to set its **Mass** to 200 g. Set the **Water mass** to 200 g. Check that the **Water temp** is set to 30.0 °C and the copper's **Temp** is 90 °C. Select the **GRAPH** tab, and click **Play** (▶).

- A. What was the **Final temperature** of the copper and the water?
- B. How much did the temperature of the copper change?
- C. How much did the temperature of the water change?

34.96

down 55.04

down 165.04

2. Specific heat capacity can be described as a substance's resistance to temperature changes. Which substance has a greater specific heat capacity, copper or water? Explain.

# Calorimetry Lab Answers

**Wolfgang Guggemos**



## **Calorimetry Lab Answers:**

## Enjoying the Melody of Appearance: An Mental Symphony within **Calorimetry Lab Answers**

In a world consumed by monitors and the ceaseless chatter of quick interaction, the melodic splendor and emotional symphony developed by the published term usually disappear in to the backdrop, eclipsed by the constant sound and interruptions that permeate our lives. But, situated within the pages of **Calorimetry Lab Answers** a wonderful fictional treasure filled with organic emotions, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this interesting masterpiece conducts readers on a psychological journey, skillfully unraveling the concealed melodies and profound impact resonating within each cautiously constructed phrase. Within the depths with this moving assessment, we shall examine the book is key harmonies, analyze its enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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