

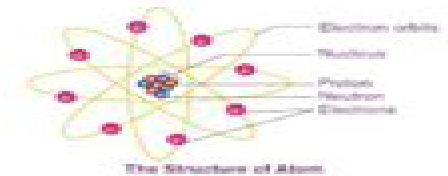
## Chapter 2: The Chemistry of Life

### ➤ Legos of the Universe

- Atoms
- Elements
  - Gold, carbon, silver, etc. (periodic table)

### ➤ Atomic Structure

- Nucleus
  - Proton (+)
  - Neutrons (no charge)
- Electron Shell



### ➤ Isotopes

- Atomic Mass

### ➤ Radioactive Atoms

- Some atomic nuclei are not stable and break down randomly
- These atoms are radioactive
- During decomposition they release, at a constant rate, a tiny high-speed particle carrying a lot of energy

### ➤ Electron shells determine bonding patterns

### ➤ Versatility of Carbon

- Can make 4 bonds
- Most often with C, H, O, and N
- Huge potential variety

### ➤ Ions

- Particles in the outer shell

### ➤ Changing Parts of the Atom

- What happens if you...
  - Add an electron to chlorine
  - Add a neutron to nitrogen
  - Add a proton to hydrogen



### ➤ Elements in Living Things

- Oxygen
- Carbon
- Hydrogen
- Nitrogen

### ➤ Molecules

- Molecules
- Bond energy
- Chemical reactions
  - **Reactants > Products**

### ➤ Chemical Bonding

- Covalent
- Ionic



# Chapter 2 The Chemistry Of Life Answer Key Concept Map

**JG Myers**



**Chapter 2 The Chemistry Of Life Answer Key Concept Map:**

## **Chapter 2 The Chemistry Of Life Answer Key Concept Map** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter 2 The Chemistry Of Life Answer Key Concept Map**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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