

# Chemistry

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(Chapter 10) (Haloalkanes and Haloarenes) (Intext Questions)  
(Class 12)

## Question 10.8:

In the following pairs of halogen compounds, which compound undergoes faster  $S_N1$  reaction?



## Answer 10.8:



$S_N1$  reaction proceeds via the formation of carbocation. The alkyl halide (I) is  $3^\circ$  while (II) is  $2^\circ$ . Therefore, (I) forms  $3^\circ$  carbocation while (II) forms  $2^\circ$  carbocation. Greater the stability of the carbocation, faster is the rate of  $S_N1$  reaction. Since  $3^\circ$  carbocation is more stable than  $2^\circ$  carbocation, (I), i.e. 2-chloro-2-methylpropane, undergoes faster  $S_N1$  reaction than (II) i.e., 3-chloropentane.

(ii)



The alkyl halide (I) is  $2^\circ$  while (II) is  $1^\circ$ .  $2^\circ$  carbocation is more stable than  $1^\circ$  carbocation. Therefore, (I), 2-chloroheptane, undergoes faster  $S_N1$  reaction than (II), 1-chloroheptane.

## Question 10.9:

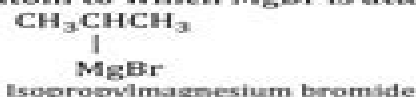
Identify A, B, C, D, E, R and R<sup>1</sup> in the following:



## Answer 10.9:



Since D of D<sub>2</sub>O gets attached to the carbon atom to which MgBr is attached, C is



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# Chemistry Chapter 10

**Richard Bailey**



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## **Table of Contents Chemistry Chapter 10**

1. Understanding the eBook Chemistry Chapter 10
  - The Rise of Digital Reading Chemistry Chapter 10
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 10
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 10
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 10
  - Personalized Recommendations
  - Chemistry Chapter 10 User Reviews and Ratings
  - Chemistry Chapter 10 and Bestseller Lists
5. Accessing Chemistry Chapter 10 Free and Paid eBooks
  - Chemistry Chapter 10 Public Domain eBooks
  - Chemistry Chapter 10 eBook Subscription Services

- Chemistry Chapter 10 Budget-Friendly Options
- 6. Navigating Chemistry Chapter 10 eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Chapter 10 Compatibility with Devices
  - Chemistry Chapter 10 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 10
  - Highlighting and Note-Taking Chemistry Chapter 10
  - Interactive Elements Chemistry Chapter 10
- 8. Staying Engaged with Chemistry Chapter 10
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Chapter 10
- 9. Balancing eBooks and Physical Books Chemistry Chapter 10
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 10
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter 10
  - Setting Reading Goals Chemistry Chapter 10
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter 10
  - Fact-Checking eBook Content of Chemistry Chapter 10
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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