

## **CHM 101: General Chemistry I (2,Units)**

### **Learning Outcomes**

At the end of this course, the students should be able to:

1. define atom, molecules and chemical reactions;
2. discuss the Modern electronic theory of atoms;
3. write electronic configurations of elements on the periodic table;
4. justify the trends of atomic radii, ionization energies, electronegativity of the elements based on their position in the periodic table;
5. identify and balance oxidation – reduction equation and solve redox titration problems;
6. illustrate shapes of simple molecules and hybridized orbitals;
7. identify the characteristics of acids, bases and salts, and solve problems based on their quantitative relationship;
8. apply the principles of equilibrium to aqueous systems using LeChatelier's principle to predict the effect of concentration, pressure and temperature changes on equilibrium mixtures;
9. analyse and perform calculations with the thermodynamic functions, enthalpy, entropy and free energy; and
10. determine rates of reactions and its dependence on concentration, time and temperature.

### **Course Contents**

Atoms, molecules, elements and compounds and chemical reactions. Modern electronic theory of atoms. Electronic configuration, periodicity and building up of the periodic table. Hybridization and shapes of simple molecules. Valence Forces and structure of solids. Chemical equations and stoichiometry, chemical bonding and intermolecular forces, kinetic theory of matter. Elementary thermochemistry, rates of reaction, equilibrium and thermodynamics. Acids, bases and salts. Properties of gases. Redox reactions and introduction to electrochemistry. Radioactivity.

# Chm 101 Noun Course Material

**Thirumalaisamy P.  
Velavancorresponding**

**Chm 101 Noun Course Material:**

## Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Chm 101 Noun Course Material**

In a world inundated with displays and the cacophony of instant interaction, the profound power and psychological resonance of verbal art often fade into obscurity, eclipsed by the regular assault of sound and distractions. Yet, nestled within the lyrical pages of **Chm 101 Noun Course Material**, a charming work of fictional beauty that pulses with natural thoughts, lies an remarkable trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, this interesting opus books viewers on a psychological odyssey, delicately exposing the latent potential and profound affect stuck within the delicate internet of language. Within the heart-wrenching expanse with this evocative analysis, we shall embark upon an introspective exploration of the book is central themes, dissect its fascinating writing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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