

CHAPTER - 2

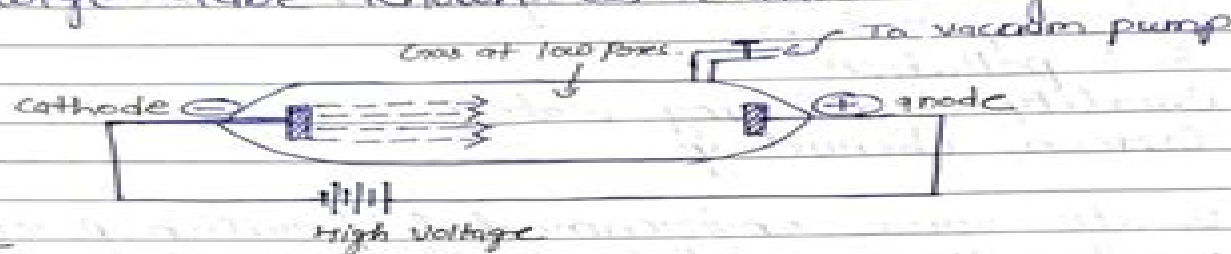
STRUCTURE OF ATOM

Discovery of electron (Cathode Rays)

Cathode rays \rightarrow William Crookes (1879)

Electron (e^-) \rightarrow J. J. Thomson (-ve charge particle)

-ve charged rays which move from cathode to anode in discharge tube known as cathode rays.



properties

- 1 Cathode rays travel in a straight line (shadow)
- 2 Cathode rays carry -ve charge (electrostatic deflection)
- 3 Cathode rays are deflected by magnetic field (rotate paddle wheel)
- 4 They are made up of material particle
- 5 They produce heating effect & possess penetrating effect
- 6 They produce green fluorescence & X-rays (when they hit the anode)
- 7 They also affect photographic plate.

Discovery of proton (Anode / Canal Rays)

Anode rays \rightarrow Goldstein (1886)

Proton (H^+) \rightarrow Goldstein & named by Rutherford (+ve charge particle)

+ve charged particle which move opposite the cathode in discharge tube known as anode rays.

Class 11th Notes Chemistry Chapter Wise

Annelies Wilder-Smith



Class 11th Notes Chemistry Chapter Wise:

Class 11th Notes Chemistry Chapter Wise Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Class 11th Notes Chemistry Chapter Wise**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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