

#### 4. pH of Prepared Buffer Solutions

Buffer Solution	Acetic Acid; 4.5%, mL	Sodium Acetate Solution, mL	Recorded pH
A	5 mL	5 mL	4.7
B	5 mL	1 mL	3.9
C	10 mL	1 mL	3.7
D	1 mL	10 mL	5.7
E	1 mL	5 mL	5.3

### III. Calculations/Results

\*show all calculations with units

1. Sodium acetate moles \_\_\_\_\_  
(molar mass = 82.0 g/mol)
2. Molarity of sodium acetate = \_\_\_\_\_
3. Concentration of acetic acid (from bottle) = \_\_\_\_\_
4. Molarity of acetic acid = \_\_\_\_\_  
(molar mass = 60.0 g/mol; density of the solution is 1.0 g/mL)

# Buffer Solution Lab Report

**Siddappa N.Byrareddy**



## **Buffer Solution Lab Report:**

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