

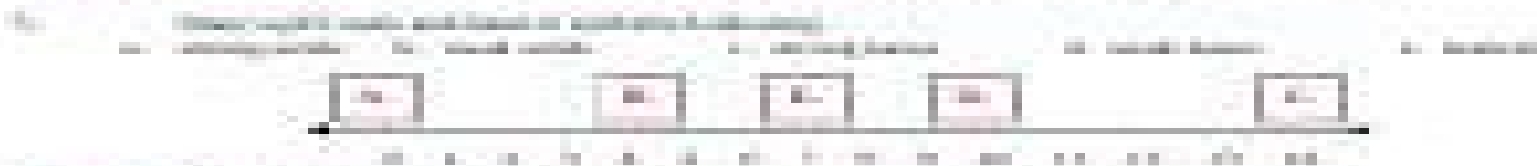
Bestimmung der Löslichkeit von Natriumacetat in Wasser bei verschiedenen Temperaturen

- 1) **Ziel:** Bestimmung der Löslichkeit von Natriumacetat in Wasser bei verschiedenen Temperaturen.
- 2) **Theorie:** Die Löslichkeit eines Festkörpers in einem flüssigen Lösungsmittel ist eine Funktion der Temperatur.
- 3) **Prinzip:** Ein bekanntes Gewicht an Natriumacetat wird in einem bestimmten Volumen Wasser gelöst. Die Temperatur wird variiert, bis sich ein gesättigtes System eingestellt hat. Die überschüssige Substanz wird abfiltriert und gewogen.
- 4) **Geräte:** Waage, Reagenzglas, Becherglas, Wasserbad, Filterpapier, Filtertrichter, Natriumacetat, Wasser.
- 5) **Reaktion:** $NaCH_3COO + H_2O \rightleftharpoons Na^+ + CH_3COO^-$

1a) **Geräte:** Waage, Reagenzglas, Becherglas, Wasserbad, Filterpapier, Filtertrichter, Natriumacetat, Wasser.

1b) **Reaktion:** $NaCH_3COO + H_2O \rightleftharpoons Na^+ + CH_3COO^-$

2) **Prinzip:** Ein bekanntes Gewicht an Natriumacetat wird in einem bestimmten Volumen Wasser gelöst. Die Temperatur wird variiert, bis sich ein gesättigtes System eingestellt hat. Die überschüssige Substanz wird abfiltriert und gewogen.



2b) **Prinzip:** Ein bekanntes Gewicht an Natriumacetat wird in einem bestimmten Volumen Wasser gelöst. Die Temperatur wird variiert, bis sich ein gesättigtes System eingestellt hat. Die überschüssige Substanz wird abfiltriert und gewogen.

3) **Reaktion:** $NaCH_3COO + H_2O \rightleftharpoons Na^+ + CH_3COO^-$

4) **Geräte:** Waage, Reagenzglas, Becherglas, Wasserbad, Filterpapier, Filtertrichter, Natriumacetat, Wasser.

5) **Reaktion:** $NaCH_3COO + H_2O \rightleftharpoons Na^+ + CH_3COO^-$

6) **Prinzip:** Ein bekanntes Gewicht an Natriumacetat wird in einem bestimmten Volumen Wasser gelöst. Die Temperatur wird variiert, bis sich ein gesättigtes System eingestellt hat. Die überschüssige Substanz wird abfiltriert und gewogen.

7) **Prinzip:** Ein bekanntes Gewicht an Natriumacetat wird in einem bestimmten Volumen Wasser gelöst. Die Temperatur wird variiert, bis sich ein gesättigtes System eingestellt hat. Die überschüssige Substanz wird abfiltriert und gewogen.

- 8a) **Geräte:** Waage, Reagenzglas, Becherglas, Wasserbad, Filterpapier, Filtertrichter, Natriumacetat, Wasser.
- 8b) **Reaktion:** $NaCH_3COO + H_2O \rightleftharpoons Na^+ + CH_3COO^-$
- 8c) **Prinzip:** Ein bekanntes Gewicht an Natriumacetat wird in einem bestimmten Volumen Wasser gelöst. Die Temperatur wird variiert, bis sich ein gesättigtes System eingestellt hat. Die überschüssige Substanz wird abfiltriert und gewogen.

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