

SECTION A (30 marks)

- A _____ is the smallest unit of an element.
A. electron C. proton
B. atom D. neutron
- Nitrogen has an atomic number of 7 and a mass number of 14. Which of the following illustrations is **CORRECT** about Nitrogen?
A. $\frac{14}{7}\text{N}$ B. $\frac{7}{14}\text{N}$ C. $\frac{7}{14}\text{N}$ D. $\frac{14}{14}\text{N}$
- The elements below are metals. Which one is **NOT**?
A. Lithium C. Beryllium
B. Helium D. Sodium
- Grade 9 learners at Tusome Junior School listed the following elements during a Science lesson.
 - Lithium
 - Oxygen
 - Potassium
 - Sodium
 Which of the above elements does not have one electron in the outermost energy level of their electron configuration?
A. Lithium C. Potassium
B. Oxygen D. Sodium
- _____ is the property of metals associated with their ability to be hammered into thin sheets without breaking.
A. Elasticity C. Ductility
B. Malleability D. Viscosity
- Brass is an alloy that is formed by combining _____ and _____.
A. Copper and Zinc C. Copper and tin
B. Iron and Carbon D. Aluminium and tin
- Iron reacts with oxygen to form _____.
A. Iron (I) oxide C. Iron (III) oxide
B. Iron (II) oxide D. Iron (IV) oxide
- We should take care of our teeth by _____.
A. eating sugary foods
B. eating hard foods
C. eating cold foods
D. eating hot foods

Learners at Gakumbwa Junior School set up the experiment below to investigate the causes of rusting in Iron. The experiment was left to stay for three days. Study it and answer 7 to 9.



- Which test tubes did not experience rusting?
A. Test tubes A and B C. Test tubes B and E
B. Test tubes A and E D. Test tubes C and D
- Which test tube experienced the highest level of rusting?
A. Test tube B C. Test tube D
B. Test tube C D. Test tube E
- During a class on functions of the parts of the male reproductive system, learners identified functions of various parts of the system. The function of testes is to _____.
A. produce hormones and sperms.
B. transport sperms only.
C. produce sperms only.
D. transport sperms and produce hormones.
- What is diffusion?
A. the movement of particles from an area of low concentration to an area of high concentration.
B. the movement of particles from an area of high concentration to an area of low concentration until equilibrium is reached.
C. the movement of particles from an area of high concentration to an area of high concentration until equilibrium is reached.
D. the movement of particles from an area of low concentration to an area of low concentration until equilibrium is reached.
- Which one is **NOT** a magnetic material?
A. Iron
B. Steel
C. Plastic
D. Nickel

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Mark Steyvers



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Table of Contents Integrated Science Midterm Exam Study Pbworks

1. Understanding the eBook Integrated Science Midterm Exam Study Pbwor
ks
 - The Rise of Digital Reading Integrated Science Midterm Exam Study Pbwor
ks
 - Advantages of eBooks Over Traditional Books
2. Identifying Integrated Science Midterm Exam Study Pbwor
ks
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Integrated Science Midterm Exam Study Pbwor
ks
 - User-Friendly Interface
4. Exploring eBook Recommendations from Integrated Science Midterm Exam Study Pbwor
ks
 - Personalized Recommendations
 - Integrated Science Midterm Exam Study Pbwor
ks User Reviews and Ratings
 - Integrated Science Midterm Exam Study Pbwor
ks and Bestseller Lists
5. Accessing Integrated Science Midterm Exam Study Pbwor
ks Free and Paid eBooks
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 - Highlighting and Note-Taking Integrated Science Midterm Exam Study Pbwor
 - Interactive Elements Integrated Science Midterm Exam Study Pbwor
- 8. Staying Engaged with Integrated Science Midterm Exam Study Pbwor
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Integrated Science Midterm Exam Study Pbwor
- 9. Balancing eBooks and Physical Books Integrated Science Midterm Exam Study Pbwor
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Integrated Science Midterm Exam Study Pbwor
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Integrated Science Midterm Exam Study Pbwor
 - Setting Reading Goals Integrated Science Midterm Exam Study Pbwor
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Integrated Science Midterm Exam Study Pbwor
 - Fact-Checking eBook Content of Integrated Science Midterm Exam Study Pbwor
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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