

**Chapter 1**  
**Matter**

# Properties of Matter

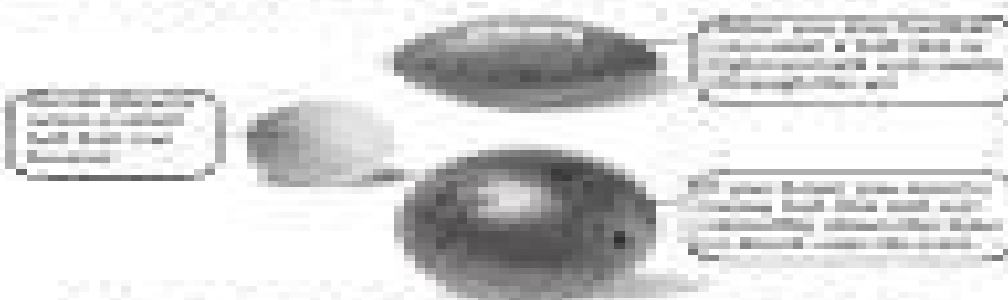
**QUESTIONS**

Are you able to distinguish, based on the questions in school

- Why are solids, liquids and gasses grouped together?
- Why are substances and mixtures classified together?

**What are Physical Properties?**

Start with the table in the chapter below as you go to understand the meaning of these common properties.



A physical property is a characteristic that can be observed without changing the identity of the substance. Most of the properties of an object can only be observed. For example, you can observe an object's color, texture, shape, size, or weight. In general, the properties of an object are determined by the particles of the substance. The object consists of **it**.

Many of the physical properties of materials that make up objects can be measured. These properties include density, hardness, magnetism, and the ability to conduct heat and electricity.

Some physical properties depend on how much of a material you have. In other words, a large amount of a material may have different properties than a smaller amount of the material. Other physical properties do not depend on how much of the material you have. They show that no matter how large or small a sample of material is, the quality of the property is always the same.

**LEARNING OBJECTIVE**

Identify the physical and chemical properties of substances and mixtures based on their physical and chemical properties.

**LEARNING OBJECTIVE**

Identify the physical and chemical properties of substances.

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**LEARNING OBJECTIVE**

Identify the physical and chemical properties of substances.

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# Chapter 2 Properties Of Matter Wordwise Answer Key

**A Gutmann**



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