



# Chapter 7 Transparency Wildgooseco

**LP Steffe**



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molecule is one where the charge distribution around the molecule is not symmetric it results from having polar bonds and  
also a molecular structure where the bond polarities do not cancel

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unequally non polar polar unequally polar the degree of polarity of a bond between any two atoms is determined by  
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bonds including covalent ionic and hydrogen bonds and london dispersion forces introduction living things are made up of  
atoms but in most cases those atoms aren t just floating around individually

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molecules and ions contain polar bonds which of these molecules and ions have dipole moments a clf 5 b clo 2 clo 2 c tecl2 4  
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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the  
molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry

such as bf 3 trigonal planar ch 4 tetrahedral pf 5 trigonal bipyramidal and sf 6

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