

CHAPTER 8 REVIEW

Chemical Equations and Reactions

SHORT ANSWER Answer the following questions in the space provided.

2. _____ According to the law of conservation of mass, the total mass of the reacting substances is
- always more than the total mass of the products.
 - always less than the total mass of the products.
 - sometimes more and sometimes less than the total mass of the products.
 - always equal to the total mass of the products.**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 2. In the reaction described by the word equation

$$\text{sodium} + \text{water} \rightarrow \text{sodium hydroxide} + \text{hydrogen gas}$$
 - sodium hydroxide is a product.
 - sodium is a product.
 - water is a reactant.
 - Both (a) and (c).**
- _____ 3. Which of the following indicates that a chemical equation is balanced?
 - The numbers of atoms of each element are the same on both sides of the equation.**
 - All of the coefficients are the same.
 - The numbers of molecules on each side of the equation are equal.
 - The sums of the coefficients on each side of the equation are equal.
- _____ 4. From a complete and correctly written chemical equation, you can obtain the
 - chemical formulas of the reactants and products.
 - relative amounts of the reactants and products.
 - physical states of the reactants and products.
 - All of the above.**
- _____ 5. To balance a chemical equation, you must
 - adjust the subscripts.
 - adjust the number of products.
 - adjust the number of reactants.
 - adjust the coefficients.**
- _____ 6. The formation of a solid product in a chemical reaction is represented by the symbol
 - (aq).
 - (g).
 - (l).
 - (s).**
- _____ 8. In the expression 3CO_2 , the numbers 3 and 2 are, respectively,
 - a subscript and a coefficient.
 - a coefficient and a subscript.**
 - two subscripts.
 - two coefficients.
- _____ 9. What is the total number of molecules that is represented by the following chemical equation? (Add up the coefficients.)

$$\text{NH}_4\text{NO}_2 \rightarrow \text{N}_2 + 2\text{H}_2\text{O}$$
 - 3
 - 4**
 - 8
 - 16
- _____ 10. Which equation below violates the law of conservation of mass?
 - $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - $\text{KCl} + \text{Br} \rightarrow \text{KBr} + \text{Cl}_2$**
 - $2\text{Fe}_2\text{O}_3 + 3\text{C} \rightarrow 4\text{Fe} + 3\text{CO}_2$
 - $\text{Na}_2\text{CO}_3 + 2\text{HCl} \rightarrow 2\text{NaCl} + \text{H}_2\text{O} + \text{CO}_2$
- _____ 11. Which of the following equations represents the balanced equation for the reaction of iron and oxygen?
 - $2\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
 - $\text{Fe} + 3\text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
 - $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$**
 - $3\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
- _____ 12. Which coefficients correctly balance the formula equation below?

$$\text{KClO}_3(\text{s}) \rightarrow \text{KCl}(\text{s}) + \text{O}_2(\text{g})$$
 - 1, 1, 1
 - 1, 1, 3
 - 2, 2, 3**
 - 2, 1, 1
- _____ 13. Which equation is *not* balanced?
 - $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - $4\text{H}_2 + 2\text{O}_2 \rightarrow 4\text{H}_2\text{O}$
 - $\text{H}_2 + \text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{H}_2\text{O}$
 - $2\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$**
- _____ 14. In which type of reaction do two or more compounds react to form one product?
 - synthesis (AKA combination)**
 - decomposition
 - combustion
 - single-replacement
- _____ 15. Which of the following is true regarding single-replacement reactions?
 - Only one product is formed.
 - There is only one reagent.
 - It requires oxygen.
 - One element replaces another in a compound.**

Chapter 8 Chemistry Answers

Puja Mehta

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Organic Synthesis** Michael Smith, 2024-08-13 Organic Synthesis 5e provides a reaction based approach to this important branch of organic chemistry Updated and accessible this eagerly awaited revision offers a comprehensive foundation for graduate students coming from disparate backgrounds and knowledge levels to provide them with critical working knowledge of basic reactions stereochemistry and conformational

principles This reliable resource uniquely incorporates molecular modeling content problems and visualizations and includes reaction examples and homework problems drawn from the latest in the current literature There have been advancements in organic reactions particularly organometallic reactions and there is a need to show how these advancements have influenced current organic synthesis The goal is to revise and update the examples of reaction examples taken from the synthesis literature from about 2017 2023 The reactions illustrate those that are used most often in modern organic synthesis but recent examples will show their current relevance Where new approaches and new reactions have been developed for organic synthesis examples will be added as new material Provides new content reaction examples and study problems from recent research Features improved organization new art and new chapter content on process chemistry and green organic chemistry Includes revised homework for each chapter with new examples and questions

CUET PG Nanoscience (Question Bank 2400 MCQs Chapterwise) Mocktime Publication, CUET PG Nanoscience Question Bank 2400 MCQs Chapterwise

Diploma in pharmacy Exit Exam Book Mr. Chandan Kumar Raikwar, Mr. Sachin Kumar Agrahari, Dr. Ravi Dnyandeo, Ms. Sandhya Ramesh Kolekar, Ms. Nikita Bibhishan Durande, 2024-10-14 This comprehensive MCQ book is meticulously designed to assist Diploma in Pharmacy students in their preparation for the Exit Exam The book covers all essential subjects ensuring a thorough understanding of key concepts Each chapter includes a diverse set of multiple choice questions providing students with a robust tool to assess their knowledge and readiness for the exam

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Study Guide, Answer Book to Accompany Organic Chemistry Alan S. Wingrove, Robert L. Caret, 1981 British Museum Catalogue of printed Books, 1897

Journal of the Franklin Institute Franklin Institute (Philadelphia, Pa.), 1956 Vols 1-69 include more or less complete patent reports of the U S Patent Office for years 1825-1859 cf Index to v 1-120 of the Journal p 415

Answer Book to Accompany Boikess/Edelson, Chemical Principles, Third Edition Robert S. Boikess, 1985

McGraw-Hill's New MCAT George J. Hademenos, 2007 This title is more focused and student friendly than competing MCAT guides It has more special features including a companion CD with two full length timed practice MCATs and free website support There is more bang for the buck this is a complete all in one preparation package for the 60 000 students who take the MCAT every year

General Science Quarterly, 1924

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Study Guide with Solutions for Organic Chemistry, Third Edition Ralph J. Fessenden, Joan S. Fessenden, 1986 Provides a set of additional drill problems chapter by chapter discussions and supplemental instructional material to help students master organic chemistry problem solving techniques

MCAT Comprehensive Review, 2004

Science Explorer Physical Science Michael J. Padilla, Ioannis

Miaculis, Martha Cyr, *Preparation for the Upper Division* Kathleen A. Hart, 1994

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