

## CHAPTER 9 REVIEW

**Stoichiometry****SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

- \_\_\_\_\_ The coefficients in a chemical equation represent the
  - masses in grams of all reactants and products.
  - relative number of moles of reactants and products.
  - number of atoms of each element in each compound in a reaction.
  - number of valence electrons involved in a reaction.
- \_\_\_\_\_ Which of the following would not be studied within the topic of stoichiometry?
  - the mole ratio of Al to Cl in the compound aluminum chloride
  - the mass of carbon produced when a known mass of sucrose decomposes
  - the number of moles of hydrogen that will react with a known quantity of oxygen
  - the amount of energy required to break the ionic bonds in  $\text{CaF}_2$
- \_\_\_\_\_ A balanced chemical equation allows you to determine the
  - mole ratio of any two substances in the reaction.
  - energy released in the reaction.
  - electron configuration of all elements in the reaction.
  - reaction mechanism involved in the reaction.
- \_\_\_\_\_ The relative number of moles of hydrogen to moles of oxygen that react to form water represents a(n)
  - reaction sequence.
  - bond energy.
  - mole ratio.
  - element proportion.
- Given the reaction represented by the following unbalanced equation:  
 $\text{N}_2\text{O}(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{NO}_2(\text{g})$ 
  - Balance the equation.

\_\_\_\_\_

b. What is the mole ratio of  $\text{NO}_2$  to  $\text{O}_2$ ?

\_\_\_\_\_

c. If 20.0 mol of  $\text{NO}_2$  form, how many moles of  $\text{O}_2$  must have been consumed?

\_\_\_\_\_

d. Twice as many moles of  $\text{NO}_2$  form as moles of  $\text{N}_2\text{O}$  are consumed. True or False?

\_\_\_\_\_

e. Twice as many grams of  $\text{NO}_2$  form as grams of  $\text{N}_2\text{O}$  are consumed. True or False?

# Chapter 9 Stoichiometry Answers Section 2

**John M. DeKorte**



## Chapter 9 Stoichiometry Answers Section 2:

**Chemistry Workbook For Dummies with Online Practice** Chris Hren, Peter J. Mikulecky, 2017-04-17 Take the confusion out of chemistry with hundreds of practice problems Chemistry Workbook For Dummies is your ultimate companion for introductory chemistry at the high school or college level Packed with hundreds of practice problems this workbook gives you the practice you need to internalize the essential concepts that form the foundations of chemistry From matter and molecules to moles and measurements these problems cover the full spectrum of topics you ll see in class and each section includes key concept review and full explanations for every problem to quickly get you on the right track This new third edition includes access to an online test bank where you ll find bonus chapter quizzes to help you test your understanding and pinpoint areas in need of review Whether you re preparing for an exam or seeking a start to finish study aid this workbook is your ticket to acing basic chemistry Chemistry problems can look intimidating it s a whole new language with different rules new symbols and complex concepts The good news is that practice makes perfect and this book provides plenty of it with easy to understand coaching every step of the way Delve deep into the parts of the periodic table Get comfortable with units scientific notation and chemical equations Work with states phases energy and charges Master nomenclature acids bases titrations redox reactions and more Understanding introductory chemistry is critical for your success in all science classes to follow keeping up with the material now makes life much easier down the education road Chemistry Workbook For Dummies gives you the practice you need to succeed *MCAT Comprehensive Review 1999*, 1998-11 Complete Review of ALL the science you need to know Full Length Practice Test with complete explanations for each question 2 Full Length Verbal Reasoning Tests with detailed explanations and strategies **ATTRACTION** Penny Reid ,2015-04-09 He is everything she doesn t want so why does she want him so badly From the New York Times Bestselling Author Penny Reid One week Private beach Invisible girl Jerk faced bully What s the worst that could happen Kaitlyn Parker has no problem being the invisible girl which is why she finds herself hiding in various cabinets and closets all over her college campus Despite her best efforts she can t escape the notice of Martin Sandeke bad boy jerkface bully and the universe s hottest wealthiest and most unobtainable bachelor who also happens to be Kaitlyn s chemistry lab partner Kaitlyn might be the only girl who isn t interested in exploiting his stunning rower s build chiseled features and family s billionaire fortune Kaitlyn wants Martin for his brain specifically to tabulate findings of trace elements in surface water When Kaitlyn saves Martin from a nefarious plot Martin uses the opportunity to push Kaitlyn out of her comfort zone spring break one week house parties bathing suits and suntan lotion Can she overcome her aversion to being noticed Will he be able grow beyond his self centered nature Or despite their obvious chemistry will Martin be the one to drive Kaitlyn into the science cabinet of obscurity for good Elements of Chemistry ATTRACTION is the first part in a three part series it is 45k words and it ends with a cliffhanger Part 1 ATTRACTION Available Now ends on a cliffhanger Part 2 HEAT Available Now ends on a

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