



The Chemistry of Life

Chapter Test A

Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

1. If an atom has a mass number of 23 and 11 electrons, the atom is most likely to
 - a. be an ion
 - b. be a cation
 - c. be a neutral atom
 - d. be an anion
2. If an atom has a mass number of 23 and 11 protons, the atom is most likely to
 - a. be an ion
 - b. be a cation
 - c. be a neutral atom
 - d. be an anion
3. Which of the following is not a characteristic of water?
 - a. low density of hydrogen-bonded water molecules
 - b. very strong surface tension and surface adhesion
 - c. moderate solubility and high capacity of hydrogen
 - d. moderate solubility and low density of water
4. A molecule that is neutral in the standard
 - a. bonding condition
 - b. bonding process
 - c. bonding electron
 - d. bonding process
5. Which molecule is polar, with the
 - a. oxygen with being slightly positive and the hydrogen with being slightly negative
 - b. oxygen with hydrogen with being slightly positive
 - c. oxygen with hydrogen with being slightly negative
 - d. oxygen with being slightly negative and the hydrogen with being slightly positive
6. If you are with water, which is
 - a. positive with negative
 - b. positive with negative
 - c. positive with positive
 - d. positive with
7. A substance with a pH of 7 is called
 - a. an acid
 - b. basic
 - c. acidic
 - d. a compound
8. Sodium chloride has a pH of 7. It is
 - a. neutral
 - b. basic
 - c. acidic
 - d. neutral
9. Why is coffee considered compared to other drinks?
 - a. Coffee does not contain any acids and therefore has a different structure.
 - b. Coffee does not have quality taste.
 - c. With coffee does not contain molecules with oxygen and hydrogen.
 - d. With coffee does not the structure of water molecules and compounds.

Chapter Test Chemistry Of Life Answer Key

David G. Myers



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