

Stoichiometry Review

Directions: Each of the problems below reviews a different type of stoichiometry. Be sure to read the problem carefully to know what it is asking for.

Mass-to-Mass Problems

1. How many grams of potassium chloride, KCl, are produced during the decomposition of 100.0 grams of potassium chlorate, KClO_3 ?



$$\frac{100 \text{ g KClO}_3}{1} \times \frac{1 \text{ mol KClO}_3}{122.545 \text{ g KClO}_3} \times \frac{1 \text{ mol KCl}}{1 \text{ mol KClO}_3} \times \frac{74.548 \text{ g KCl}}{1 \text{ mol KCl}} = \underline{60.83 \text{ g KCl}}$$

2. Find the number of grams of HCl needed to react completely with 0.50 moles of magnesium.



$$\frac{0.5 \text{ mol Mg}}{1} \times \frac{1 \text{ mol HCl}}{1 \text{ mol Mg}} \times \frac{36.458 \text{ g HCl}}{1 \text{ mol HCl}} = \underline{18.23 \text{ g HCl}}$$

3. Find the mass in grams of hydrogen gas produced when 14.0 moles of HCl are added to an excess amount of magnesium.



$$\frac{14 \text{ mol HCl}}{1} \times \frac{1 \text{ mol H}_2}{2 \text{ mol HCl}} \times \frac{2.016 \text{ g H}_2}{1 \text{ mol H}_2} = \underline{14.112 \text{ g H}_2}$$

4. If 0.75 moles of ammonia are needed, how many grams of nitrogen will be consumed?



$$\frac{0.75 \text{ mol NH}_3}{1} \times \frac{1 \text{ mol N}_2}{2 \text{ mol NH}_3} \times \frac{28.014 \text{ g N}_2}{1 \text{ mol N}_2} = \underline{10.51 \text{ g N}_2}$$

Review Stoichiometry Section 1 And 2 Answers

Yicheng Fang



Review Stoichiometry Section 1 And 2 Answers:

Adopting the Track of Phrase: An Psychological Symphony within **Review Stoichiometry Section 1 And 2 Answers**

In a world used by displays and the ceaseless chatter of instant interaction, the melodic elegance and psychological symphony created by the prepared term often fade in to the background, eclipsed by the persistent sound and interruptions that permeate our lives. Nevertheless, set within the pages of **Review Stoichiometry Section 1 And 2 Answers** an enchanting fictional value brimming with organic emotions, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, this fascinating masterpiece conducts visitors on a mental trip, well unraveling the hidden tunes and profound influence resonating within each carefully crafted phrase. Within the depths of this touching review, we will examine the book is key harmonies, analyze its enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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