

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ a. CH_3COOH
- _____ b. Lewis Acids
- _____ c. Carbonyl Compounds
- _____ d. Alcohols
- _____ e. Metal Ions
- _____ f. Simple Molecules
- _____ g. Methyl Radicals
- _____ h. Lewis Bases
- _____ i. Halogenated Ions
- _____ j. H^+ Ions

- _____ k. bonded by coordinate covalent bonds that are superimposed on other bonds
- _____ l. single group of negatively charged atoms
- _____ m. the energy released when one mole of an ionic crystalline compound is decomposed into
- _____ n. cations or anions
- _____ o. coordinate covalent bonds are part of structure in which both atoms
- _____ p. the energy required to break a chemical bond and form gaseous ions
- _____ q. structural group of atoms that are held together by covalent bonds, results from the sharing of electron pairs between atoms
- _____ r. chemical bonding that results from the electrical attraction between positive and negative
- _____ s. mutual electrical attraction between oppositely charged ions, and the resulting chemical bond, particularly the ionic compound

Multiple Choice

2. The charge on the ion is _____

- _____ a. always positive
- _____ b. always negative
- _____ c. either positive or negative
- _____ d. neither

3. In many compounds, atoms of one group form a bond with atoms of another of another of the same compound, forming bonds of both atoms _____

- _____ a. 2
- _____ b. 3
- _____ c. 4
- _____ d. 5

4. Which of the following is not a Lewis acid? _____

- _____ a. Lewis Acids
- _____ b. Carbonyl Compounds
- _____ c. Alcohols
- _____ d. Metal Ions
- _____ e. Simple Molecules
- _____ f. Methyl Radicals
- _____ g. Lewis Bases
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- _____ i. H^+ Ions

Modern Chemistry Chapter 6 Test Answers

Frank L. Wiseman



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