

Please answer each question carefully. Remember, follow your gut! Please copy all answers on the answer sheet on the back of this exam! Otherwise, I will take off 10 points!

- The structure of DNA described by Watson and Crick included:
 - a double helix
 - the sugar-phosphate backbone aligned in the center of the helix
 - the base pairs that are stacked on the inside of the double helix
 - A double helix and base pairs that are stacked on the inside of the double helix
- What gives proteins such a dominant role in biochemistry?
 - the ability to act as a blueprint
 - their ability to spontaneously fold into complex three-dimensional structures
 - the variation in protein sizes
 - their ability to self-replicate
- Match the loss of a particular organelle with the associated disease.
 - Hypercholesterolemia - smooth endoplasmic reticulum
 - Diabetes - endosome
 - Tay-Sachs disease - lysosome
 - Muscle degeneration - mitochondria
- In a biochemistry lab course, you are asked to design an experiment to identify a strain of bacteria. Your lab partner claims that she thinks the bacterium contains a rough endoplasmic reticulum. To verify her claim, which of the following experiments would you perform? (pretend you're indulging her ludicrous claim).
 - determine whether the bacterium has an internal membrane-enclosed compartment
 - determine whether the bacterium generates CO_2 in the presence of fuel molecules
 - determine whether the bacterium can synthesize proteins
 - determine whether the bacterium can synthesize ATP in the presence of fuel molecules and O_2
- What is the H^+ concentration in a urine sample that has a pH of 6?
 - 10^{-6} M
 - 10^{-8} M
 - 10^6 M
 - 10^{-14} M
- Which of the following is considered a noncovalent bond?
 - electrostatic interactions
 - hydrogen bonds
 - van der Waals interactions
 - All of these answers
- Which functional group is shown below?



- Amine
 - Thiol
 - Carboxylic acid
 - None of these answers
- What is the $[\text{A}^-]/[\text{HA}]$ ratio when the weak acid is in a solution one pH unit above its pK_a ?

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G Orfield



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Table of Contents Biochemistry Short Answer Questions

1. Understanding the eBook Biochemistry Short Answer Questions
 - The Rise of Digital Reading Biochemistry Short Answer Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Biochemistry Short Answer Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biochemistry Short Answer Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biochemistry Short Answer Questions
 - Personalized Recommendations
 - Biochemistry Short Answer Questions User Reviews and Ratings
 - Biochemistry Short Answer Questions and Bestseller Lists
5. Accessing Biochemistry Short Answer Questions Free and Paid eBooks
 - Biochemistry Short Answer Questions Public Domain eBooks
 - Biochemistry Short Answer Questions eBook Subscription Services
 - Biochemistry Short Answer Questions Budget-Friendly Options
6. Navigating Biochemistry Short Answer Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Biochemistry Short Answer Questions Compatibility with Devices
 - Biochemistry Short Answer Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biochemistry Short Answer Questions
 - Highlighting and Note-Taking Biochemistry Short Answer Questions
 - Interactive Elements Biochemistry Short Answer Questions

8. Staying Engaged with Biochemistry Short Answer Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biochemistry Short Answer Questions
9. Balancing eBooks and Physical Books Biochemistry Short Answer Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biochemistry Short Answer Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Biochemistry Short Answer Questions
 - Setting Reading Goals Biochemistry Short Answer Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Biochemistry Short Answer Questions
 - Fact-Checking eBook Content of Biochemistry Short Answer Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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