

Covalent Bonding Review Sheet – Dec 12, 2012

Name _____ class period _____

- Given the formula, a compound can be named using conventional systems that include Greek prefixes and Roman numeral where appropriate.
- Differences in electronegativity values can be used to predict where a bond fits on the continuum between ionic and covalent bonds.
- Since most compounds contain multiple bonds, a substance may contain more than one type of bond.

Section 1: Naming and Formula Writing

Write the formulas for the following covalent compounds

1. Nitrogen Monoxide NO
2. Dinitrogen Monoxide N₂O
3. Nitrogen Dioxide NO₂
4. Dinitrogen Trioxide N₂O₃
5. Dinitrogen Tetraoxide N₂O₄
6. Dinitrogen Pentaoxide N₂O₅
7. carbon dioxide CO₂
8. phosphorus triiodide PI₃
9. sulfur dichloride SCl₂
10. nitrogen trifluoride NF₃

Write names from the following formulas

11. N₂F₄ dinitrogen tetrafluoride
12. SCl₄ sulfur tetrachloride
13. ClF₃ chlorine trifluoride
14. SiO₂ silicon dioxide
15. P₄O₁₀ tetraphosphorus decaoxide
16. SiF₄ silicon tetrafluoride
17. H₂O dihydrogen monoxide
18. Br₂ dibromide or bromine
19. HBr hydrogen monobromide
20. N₂S₃ dinitrogen trisulfide

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Chemistry3 Andrew Burrows, John Holman, Andrew Parsons, Gwen Pilling, Gareth Price, 2017 Chemistry3 establishes the fundamental principles of all three strands of chemistry organic inorganic and physical By building on what students have learned at school using carefully worded explanations annotated diagrams and worked examples it presents an approachable introduction to chemistry and its relevance to everyday life Concepts in Biology' 2007 Ed.2007 Edition , **Holt**

Chemistry Ralph Thomas Myers, 2004 **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

CHEMISTRY SILBERBERG, 2003 **Human Biology** Daniel Chiras, 2012 Written for the introductory human biology course the Seventh Edition of Chiras acclaimed text maintains the original organizational theme of homeostasis presented in previous editions to present the fundamental concepts of mammalian biology and human structure and function Chiras discusses the scientific process in a thought provoking way that asks students to become deeper more critical thinkers The focus on health and homeostasis allows students to learn key concepts while also assessing their own health needs An updated and enhanced ancillary package includes numerous student and instructor tools to help students get the most out of their course **Human Biology** Daniel D. Chiras, 2005 Intended for non majors this textbook describes the structure and functions of each human body system explores the body processes that regulate chemical levels in the blood and body temperature and overviews genetics human reproduction and evolution The fifth edition trims the overall length by 20% while adding short essays on past scientific

Chemistry Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives Biology Joseph S. Levine, Kenneth R. Miller, 1998 **Chemistry** John S. Phillips, Cheryl Wistrom, 2000 *General Science* , 1989 *Fundamentals of Chemistry* David E. Goldberg, 2001 Chemical Interactions , 2005 Biology Kenneth R. Miller, 1991

Prentice Hall Science Explorer Michael J. Padilla, 2002 Modules McDougal Littell Incorporated, 2005 *Science Books & Films* , 1982 **Biochemistry Review** David M. Glick, 1975 Previous editions by E H Mosbach **Biology** Neil A. Campbell, Jane B. Reece, 2005 Neil Campbell and Jane Reece s BIOLOGY remains unsurpassed as the most successful majors biology textbook in the world This text has invited more than 4 million students into the study of this dynamic and essential discipline The authors have restructured each chapter around a conceptual framework of five or six big ideas An Overview draws students in and sets the stage for the rest of the chapter each numbered Concept Head announces the beginning of a new concept and Concept Check questions at the end of each chapter encourage students to assess their mastery of a given concept New Inquiry Figures focus students on the experimental process and new Research Method Figures illustrate important techniques in biology Each chapter ends with a Scientific Inquiry Question that asks students to apply scientific

investigation skills to the content of the chapter *Chemical principles* Paul B. Kelter, 1998

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