

- 4 Chlorine has two isotopes,  ${}_{17}^{35}\text{Cl}$  and  ${}_{17}^{37}\text{Cl}$ .
- (a) What do you understand by the term *isotope*? [1]
- (b) Write down the number of neutrons in each of these isotopes of chlorine. [2]
- (c) Draw and label a diagram of an atom of chlorine to show:
- the nucleus
  - the electron shells, with the correct number of electrons in each shell. [3]
- (d) A chloride ion has eight electrons in its outer shell. What is the importance of this electronic configuration? [1]

*(Paper 3)*

# Chapter 2 Section Review Questions

**S Ashworth**



**Chapter 2 Section Review Questions:**

## Unveiling the Magic of Words: A Review of "**Chapter 2 Section Review Questions**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Chapter 2 Section Review Questions**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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