

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers without answers with symbols.

- The molar volume of an ideal gas at standard temperature and pressure is  $22.7 \text{ L mol}^{-1}$ .
  - gas
  - pressure
  - standard temperature
  - volume
- The volume of a gas is independent of its pressure at constant temperature.
  - constant
  - independent
  - inversely proportional
  - proportional to pressure
- An unknown substance that contains only hydrogen and the unknown is
  - water
  - aluminum hydroxide
  - acetic acid
  - methane
- The molar mass of a compound is  $100.0 \text{ g mol}^{-1}$ . The molar mass of the
  - hydrogen atoms in the compound is  $1.008 \text{ g mol}^{-1}$ .
  - oxygen atoms in the compound is  $16.00 \text{ g mol}^{-1}$ .
  - hydrogen atoms in the compound is  $1.008 \text{ g mol}^{-1}$ .
  - oxygen atoms in the compound is  $16.00 \text{ g mol}^{-1}$ .
- The molar mass of a compound is  $100.0 \text{ g mol}^{-1}$ .
  - hydrogen: the number of moles of H is unknown. The mass of H is the number of grams of H in unknown. The mass of H is the number of grams of H in unknown. The mass of H is the number of grams of H in unknown.
  - oxygen: the number of moles of O is unknown. The mass of O is the number of grams of O in unknown. The mass of O is the number of grams of O in unknown.
  - hydrogen: the number of moles of H is unknown. The mass of H is the number of grams of H in unknown. The mass of H is the number of grams of H in unknown.
  - oxygen: the number of moles of O is unknown. The mass of O is the number of grams of O in unknown. The mass of O is the number of grams of O in unknown.
- At  $25^\circ\text{C}$ , the molar volume of an ideal gas is  $24.5 \text{ L mol}^{-1}$ .
  - At  $25^\circ\text{C}$ , the molar volume of an ideal gas is  $24.5 \text{ L mol}^{-1}$ .
  - At  $25^\circ\text{C}$ , the molar volume of an ideal gas is  $24.5 \text{ L mol}^{-1}$ .
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  - At  $25^\circ\text{C}$ , the molar volume of an ideal gas is  $24.5 \text{ L mol}^{-1}$ .

# Chapter Test B Chemical Equations And Reactions Answers

**CO Houle**



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