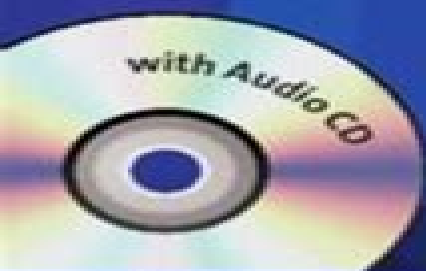


Cambridge English

Objective Proficiency

Workbook
with answers

For revised exam from March 2013



Peter Sunderland Erica Whettem

Second Edition

Official preparation material for Cambridge English: Proficiency,
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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined the first reactant in this equation contains magnesium mg and a hydroxide ion oh 1 which is a polyatomic anion

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web nov 9 2023 to balance a chemical equation follow these four simple steps to show the reactants and products write the imbalanced equation calculate the number of atoms of each element on each side of the reaction arrow to make the number of atoms of each element the same on both sides of the equation multiply coefficients the numbers in

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web oct 13 2023 this means that you will need to balance the carbon atoms first 5 use a coefficient to balance the single carbon atom add a coefficient to the single carbon atom on the right of the equation to balance it with the 3 carbon atoms on the left of the equation $\text{C}_3\text{H}_8 + 2\text{H}_2\text{O} \rightarrow 3\text{CO}_2$