

The Mole & Stoichiometry

Across

3. What branch of science was the mole used

5. a unit of amount, the amount of a substance containing 6.02×10^{23} particles

8. What is the same number as the atomic number?

9. the molar mass of an element is the same as it's

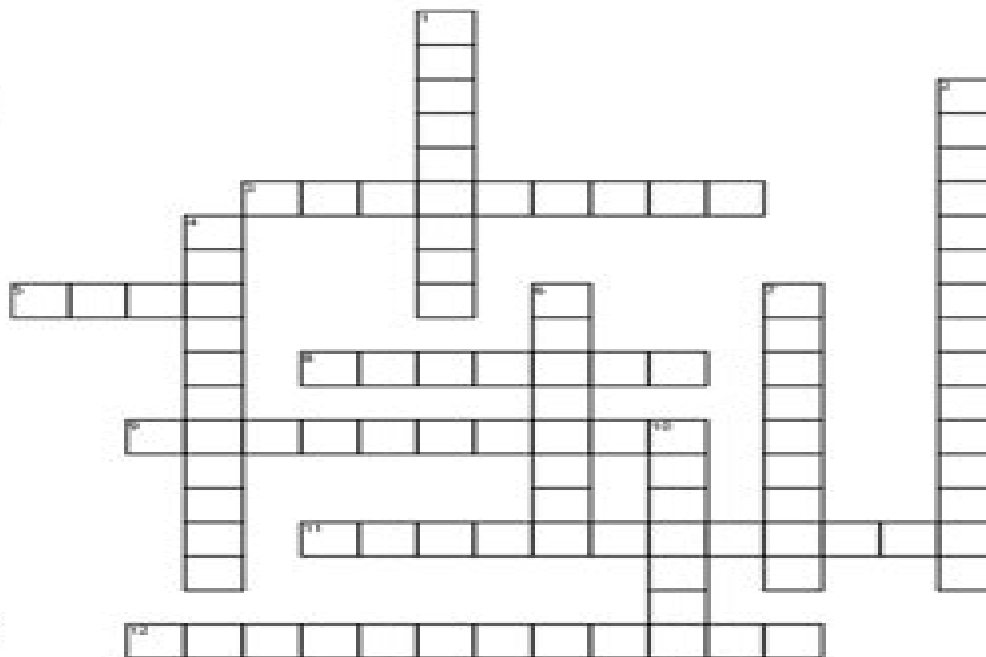
11. What serves as a standard for the mole unit

12. the numbers from a balanced chemical equation that determine mole ratios

Down

1. The mass of one mole of a substance is called the

2. what do we call the number 6×10^{23}



4. In Avogadro's number what is the 10 raised to?

6. who discovered the mole? (last name)

7. the moles measure the number of atoms, molecules or formula units all known as _____

10. What is the molar mass of Oxygen?



Stoichiometry Crossword Answers

David Thurlo



Stoichiometry Crossword Answers:

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You Can Do Chemistry Sunil Tanna, 2018-11-19 A comprehensive guide to performing mole and stoichiometric calculations with numerous examples as well as questions and answers Covers calculations relating to solids solutions gases and electrolysis plus as limiting and excess reactants chemical yields atom economy and much more Fully up to date with the last international standards including the revised definition of mole which was agreed on November 16th 2018

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Chemistry Games Gregory Gebhardt,2012-10-01 This book is a science education text It is a collection of Chemistry games that teaches the periodic table stoichiometry chemical reactions and the law of conservation of mass

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