

MOLAR CONVERSIONS LAB

Many everyday items are counted in groups in order to make them more manageable. Eggs are counted in dozens; sheets of paper are counted in reams. Since atoms and molecules are so minuscule, it is necessary to count them in groups much larger than 12 or 500. Instead, they are counted in groups called moles. The mole is a group of 6.02×10^{23} particles, or Avogadro's number (N_A) of particles.

Counting the individual items in a group as large as the mole is impossible. As a result, a different method is used to count a mole of particles – counting by weighing. The same process is used to count aluminum cans when you take a bag to the recycling center. A conversion factor (weight per can) is used to convert the weight of the bag into the number of cans in the bag. In the same way, the molar mass of an element or compound is used to convert the mass of a sample into the number of atoms or particles in that sample.

In this lab, you will perform molar conversions for small samples of two everyday substances, water and chalk.

PROCEDURE

1. Fill a paper cup with water and measure its mass.
2. Drink one mouthful of water and find the new mass of the paper cup.
3. Record the initial and final masses in your data table.
4. Obtain a piece of chalk (calcium carbonate). Wipe it free of dust particles and measure its mass.
5. Write your full name with the chalk and find its new mass.
6. Record the initial and final masses in your data table.

DATA – Molar Conversions Data Table

	Water	Chalk
Initial Mass (g)		
Final Mass (g)		
Mass consumed (g)		

ANALYSIS – Information for Each Substance (show work on the back of this sheet)

	Water	Chalk
Chemical Formula		
Molar Mass		
Particles Consumed in Lab		

QUESTIONS – Answer on the back of this sheet

1. Identify the two conversion factors you used and specify how you used them.
2. If you consume 5 grams each of water and chalk, will you consume more water particles or more chalk particles? Explain.
3. If you consume 5 moles each of water and chalk, will you consume more water particles or more chalk particles? Explain.

Moles Of Chalk Lab Answers

SJ Ball



Moles Of Chalk Lab Answers:

E-chemistry Iii (science and Technology)' 2003 Ed. , *Laboratory Manual for Principles of General Chemistry* Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures **Canadian Journal of Chemistry** ,1968 *Physics Guide and Lab Activities* Ira Cleveland Davis,1957 **Prentice Hall Chemistry** ,2000 Christian Home Educators' Curriculum Manual Cathy Duffy,1997-11 *Basic Chemistry* Steven S. Zumdahl,2004 Description Not Yet Available *Nuclear Science Abstracts* ,1970 **Illinois Chemistry Teacher** ,2003 **Government Reports Announcements & Index** ,1983 **Improving Student Comprehension of Stoichiometric Concepts** Connie Lynn Bannick Kemner,2007 *Biology* Eric Strauss,Marylin Lisowski,2000 **Canadian Chemical Processing** ,1969-07

Decoding **Moles Of Chalk Lab Answers**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Moles Of Chalk Lab Answers**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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