

Periodic Trends

ATOMIC RADIUS

1. What trend in atomic radius do you see as you go down a group/family on the periodic table?
2. What causes this trend?
3. What trend in atomic radius do you see as you go across a period/row on the periodic table?
4. What causes this trend?
5. Circle the atom in each pair that has the largest atomic radius.
 - a) Al B
 - b) S O
 - c) Br Cl
 - d) Na Al
 - e) O F
 - f) Mg Ca
6. Put the following elements in order from smallest to largest atomic radius **and** explain why:
C, O, Sn, Sr.

ELECTRONEGATIVITY

7. Define electronegativity
8. How does the ionic radius of a nonmetal compare with its atomic radius?
9. What trend in electronegativity do you see as you go down a group/family on the periodic table?
10. What causes this trend?
11. What trend in electronegativity do you see as you go across a period/row on the periodic table?
12. What causes this trend?
13. Circle the atom in each pair that has the greater electronegativity.
 - a) Ca Ga
 - b) Li O
 - c) Cl S
 - d) Br As
 - e) Ba Sr
 - f) O S

GENERAL QUESTIONS

14. Which group tends to form +1 ions? _____
15. Which group tends to form +2 ions? _____
16. Which group tends to form -1 ions? _____
17. Which group tends not to form ions or react? _____

Periodic Table Chapter Test A Answer Key

Huangqi Zhang



Periodic Table Chapter Test A Answer Key:

Reviewing **Periodic Table Chapter Test A Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Periodic Table Chapter Test A Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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Table of Contents Periodic Table Chapter Test A Answer Key

1. Understanding the eBook Periodic Table Chapter Test A Answer Key
 - The Rise of Digital Reading Periodic Table Chapter Test A Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Chapter Test A Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Chapter Test A Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Chapter Test A Answer Key
 - Personalized Recommendations
 - Periodic Table Chapter Test A Answer Key User Reviews and Ratings

- Periodic Table Chapter Test A Answer Key and Bestseller Lists
- 5. Accessing Periodic Table Chapter Test A Answer Key Free and Paid eBooks
 - Periodic Table Chapter Test A Answer Key Public Domain eBooks
 - Periodic Table Chapter Test A Answer Key eBook Subscription Services
 - Periodic Table Chapter Test A Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Chapter Test A Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Chapter Test A Answer Key Compatibility with Devices
 - Periodic Table Chapter Test A Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Chapter Test A Answer Key
 - Highlighting and Note-Taking Periodic Table Chapter Test A Answer Key
 - Interactive Elements Periodic Table Chapter Test A Answer Key
- 8. Staying Engaged with Periodic Table Chapter Test A Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Chapter Test A Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Chapter Test A Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Chapter Test A Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Chapter Test A Answer Key
 - Setting Reading Goals Periodic Table Chapter Test A Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Chapter Test A Answer Key
 - Fact-Checking eBook Content of Periodic Table Chapter Test A Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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