

QUALITATIVE ANALYSIS (CATIONS)

Group	Group reagent	Cations (Basic radicals)	Composition & colour of precipitate
1	dil HCl	Pb^{2+} (w)	$PbCl_2$; white
		Ag^+	$AgCl$; white (adv)
		Hg_2^{2+}	Hg_2Cl_2 ; white
2(A)	H_2S in presence of dil HCl (insoluble in YAS)	Pb^{2+} (w)	PbS ; Black
		Cu^{2+} (w)	CuS ; Black
		Hg^{2+}	HgS ; Black (adv)
		Bi^{3+}	Bi_2S_3 ; Black
		Co^{2+} Cd^{2+}	CoS ; yellow
2(B)	H_2S in presence of dil HCl (soluble in YAS)	As^{3+}, Sb^{3+}	As_2S_3 ; yellow, Sb_2S_3 ; orange
		Sn^{2+}, Sn^{4+}	SnS ; brown, SnS_2 ; yellow
3	NH_4OH in presence of NH_4Cl	Fe^{3+} (w)	$Fe(OH)_3$; red brown
		Al^{3+} (w)	$Al(OH)_3$; gelatinous white
		Cr^{3+}	$Cr(OH)_3$; green (adv)
4	H_2S in presence of NH_4OH & NH_4Cl	Zn^{2+} (w)	ZnS ; white
		Ni^{2+} (w)	NiS ; black
5	$(NH_4)_2CO_3$ in presence of NH_4OH <small>Mns (Bubb/line)</small>	Mn^{2+} (w)	$MnCO_3$; black, MnS ; black
		Ba^{2+} (w)	$BaCO_3$; white
		Ca^{2+} (w)	$CaCO_3$; white
		Sr^{2+}	$SrCO_3$; white
6	Na_2HPO_4 in presence of NH_4OH	Mg^{2+} (w)	$Mg(NH_4)PO_4$; white

Qualitative Analysis Of Cations Lab Report Answers

George H. Schenk, Darrell D. Ebbing



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Laboratory Manual for Principles of General Chemistry Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Qualitative Analysis and Ionic Equilibrium George H. Schenk,1990 A supplement for courses with a qualitative analysis component this lab manual contains explanations of the chemistry of metal ions and anions It includes pre lab exercises experiments and lab reports

Chemistry in the Laboratory James M. Postma,Julian L. Robert,J. Leland Hollenberg,2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

Abstracts of Papers American Chemical Society. Meeting,American Chemical Society,1956

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Qualitative Analysis and Ionic Equilibrium George H. Schenk,Darrell D. Ebbing,1985 A supplement for courses with a qualitative analysis component this lab manual contains explanations of the chemistry of metal ions and anions It includes pre lab exercises experiments and lab reports

Principles Of Qualitative Inorganic Analysis Huda Alhasan; Nadiyah,Nadiyah Alahmadi,Huda S Alhasan,2021-01-26 This book provides notes for basic laboratory experiments in qualitative analysis of cations The book introduces readers to basic methods and laboratory safety Subsequent chapters cover six groups of cations Each chapter explains important details that are required to understand how a particular analytical method works for detecting cations in samples starting from sedimentation and ending with the identification Key Features Simple reader friendly format introductory notes and summary Covers several groups of metals Appendix for handy reference with tables and references This is a useful textbook for early chemistry students and teachers as it equips the readers with sufficient information required to analyze chemical samples and deduce the presence of specific cations as part of laboratory coursework

Nonsulfide Qualitative Analysis of Cations; Separating & Identifying Representative Cations from Group A-E Judith C. Foster,1995-01-01

Qualitative Analysis and the Properties of Ions in Aqueous Solution Emil J. Slowinski,William L. Masterton,1990 This inexpensive qualitative analysis supplement offers maximum flexibility and

can accompany general chemistry texts Works well with any general chemistry text where the instructor wants more qualitative analysis in conjunction with regular class work

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Table of Contents Qualitative Analysis Of Cations Lab Report Answers

1. Understanding the eBook Qualitative Analysis Of Cations Lab Report Answers
 - The Rise of Digital Reading Qualitative Analysis Of Cations Lab Report Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Qualitative Analysis Of Cations Lab Report Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Qualitative Analysis Of Cations Lab Report Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Qualitative Analysis Of Cations Lab Report Answers
 - Personalized Recommendations
 - Qualitative Analysis Of Cations Lab Report Answers User Reviews and Ratings
 - Qualitative Analysis Of Cations Lab Report Answers and Bestseller Lists
5. Accessing Qualitative Analysis Of Cations Lab Report Answers Free and Paid eBooks
 - Qualitative Analysis Of Cations Lab Report Answers Public Domain eBooks
 - Qualitative Analysis Of Cations Lab Report Answers eBook Subscription Services
 - Qualitative Analysis Of Cations Lab Report Answers Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Qualitative Analysis Of Cations Lab Report Answers Compatibility with Devices
 - Qualitative Analysis Of Cations Lab Report Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Qualitative Analysis Of Cations Lab Report Answers
 - Highlighting and Note-Taking Qualitative Analysis Of Cations Lab Report Answers
 - Interactive Elements Qualitative Analysis Of Cations Lab Report Answers
8. Staying Engaged with Qualitative Analysis Of Cations Lab Report Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Qualitative Analysis Of Cations Lab Report Answers
9. Balancing eBooks and Physical Books Qualitative Analysis Of Cations Lab Report Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Qualitative Analysis Of Cations Lab Report Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Qualitative Analysis Of Cations Lab Report Answers
 - Setting Reading Goals Qualitative Analysis Of Cations Lab Report Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Qualitative Analysis Of Cations Lab Report Answers
 - Fact-Checking eBook Content of Qualitative Analysis Of Cations Lab Report Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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- Interactive and Gamified eBooks

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