

## Balancing Equations Worksheet

- 1) \_\_\_  $\text{Na}_3\text{PO}_4$  + \_\_\_  $\text{KOH}$   $\rightarrow$  \_\_\_  $\text{NaOH}$  + \_\_\_  $\text{K}_3\text{PO}_4$
- 2) \_\_\_  $\text{MgF}_2$  + \_\_\_  $\text{Li}_2\text{CO}_3$   $\rightarrow$  \_\_\_  $\text{MgCO}_3$  + \_\_\_  $\text{LiF}$
- 3) \_\_\_  $\text{P}_4$  + \_\_\_  $\text{O}_2$   $\rightarrow$  \_\_\_  $\text{P}_2\text{O}_3$
- 4) \_\_\_  $\text{RbNO}_3$  + \_\_\_  $\text{BeF}_2$   $\rightarrow$  \_\_\_  $\text{Be}(\text{NO}_3)_2$  + \_\_\_  $\text{RbF}$
- 5) \_\_\_  $\text{AgNO}_3$  + \_\_\_  $\text{Cu}$   $\rightarrow$  \_\_\_  $\text{Cu}(\text{NO}_3)_2$  + \_\_\_  $\text{Ag}$
- 6) \_\_\_  $\text{CF}_4$  + \_\_\_  $\text{Br}_2$   $\rightarrow$  \_\_\_  $\text{CBr}_4$  + \_\_\_  $\text{F}_2$
- 7) \_\_\_  $\text{HCN}$  + \_\_\_  $\text{CuSO}_4$   $\rightarrow$  \_\_\_  $\text{H}_2\text{SO}_4$  + \_\_\_  $\text{Cu}(\text{CN})_2$
- 8) \_\_\_  $\text{GaF}_3$  + \_\_\_  $\text{Cs}$   $\rightarrow$  \_\_\_  $\text{CsF}$  + \_\_\_  $\text{Ga}$
- 9) \_\_\_  $\text{BaS}$  + \_\_\_  $\text{PtF}_2$   $\rightarrow$  \_\_\_  $\text{BaF}_2$  + \_\_\_  $\text{PtS}$
- 10) \_\_\_  $\text{N}_2$  + \_\_\_  $\text{H}_2$   $\rightarrow$  \_\_\_  $\text{NH}_3$
- 11) \_\_\_  $\text{NaF}$  + \_\_\_  $\text{Br}_2$   $\rightarrow$  \_\_\_  $\text{NaBr}$  + \_\_\_  $\text{F}_2$
- 12) \_\_\_  $\text{Pb}(\text{OH})_2$  + \_\_\_  $\text{HCl}$   $\rightarrow$  \_\_\_  $\text{H}_2\text{O}$  + \_\_\_  $\text{PbCl}_2$
- 13) \_\_\_  $\text{AlBr}_3$  + \_\_\_  $\text{K}_2\text{SO}_4$   $\rightarrow$  \_\_\_  $\text{KBr}$  + \_\_\_  $\text{Al}_2(\text{SO}_4)_3$
- 14) \_\_\_  $\text{CH}_4$  + \_\_\_  $\text{O}_2$   $\rightarrow$  \_\_\_  $\text{CO}_2$  + \_\_\_  $\text{H}_2\text{O}$
- 15) \_\_\_  $\text{Na}_3\text{PO}_4$  + \_\_\_  $\text{CaCl}_2$   $\rightarrow$  \_\_\_  $\text{NaCl}$  + \_\_\_  $\text{Ca}_3(\text{PO}_4)_2$
- 16) \_\_\_  $\text{K}$  + \_\_\_  $\text{Cl}_2$   $\rightarrow$  \_\_\_  $\text{KCl}$
- 17) \_\_\_  $\text{Al}$  + \_\_\_  $\text{HCl}$   $\rightarrow$  \_\_\_  $\text{H}_2$  + \_\_\_  $\text{AlCl}_3$
- 18) \_\_\_  $\text{N}_2$  + \_\_\_  $\text{F}_2$   $\rightarrow$  \_\_\_  $\text{NF}_3$
- 19) \_\_\_  $\text{SO}_2$  + \_\_\_  $\text{Li}_2\text{Se}$   $\rightarrow$  \_\_\_  $\text{SSe}_2$  + \_\_\_  $\text{Li}_2\text{O}$
- 20) \_\_\_  $\text{NH}_3$  + \_\_\_  $\text{H}_2\text{SO}_4$   $\rightarrow$  \_\_\_  $(\text{NH}_4)_2\text{SO}_4$

# Balancing Equation Practice Packet Answer Sheet

**Peipei Pang**



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