

**ICSE Biology Grade X
Solution for 2022-23 Examination**

BIOLOGY

(SCIENCE PAPER - 3)

Maximum Marks: 80

Time allowed: Two hours

Answers to this Paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the question paper.

The time given at the head of this Paper is the time allowed for writing the answers.

Section - A (40 Marks)

(Attempt all questions from this section)

Question 1. Select the correct answers to the questions from the given options.
(Do not copy the questions, write the correct answer only).

- (i) The sex chromosome in a human ovum is:
- (a) X chromosome
 - (b) Y chromosome
 - (c) Both X and Y chromosomes
 - (d) Either X or Y chromosome

Answer - (a) X chromosome

Ovum contains one single X chromosome while sperm can have either X or Y chromosome.

- (ii) Which one of the following is a biodegradable waste?
- (a) Metal cans
 - (b) E-waste
 - (c) Plastic
 - (d) Flowers

Biology Paper 2 Question And Answer

David Kirk



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Table of Contents Biology Paper 2 Question And Answer

1. Understanding the eBook Biology Paper 2 Question And Answer
 - The Rise of Digital Reading Biology Paper 2 Question And Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology Paper 2 Question And Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology Paper 2 Question And Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology Paper 2 Question And Answer
 - Personalized Recommendations
 - Biology Paper 2 Question And Answer User Reviews and Ratings

- Biology Paper 2 Question And Answer and Bestseller Lists
- 5. Accessing Biology Paper 2 Question And Answer Free and Paid eBooks
 - Biology Paper 2 Question And Answer Public Domain eBooks
 - Biology Paper 2 Question And Answer eBook Subscription Services
 - Biology Paper 2 Question And Answer Budget-Friendly Options
- 6. Navigating Biology Paper 2 Question And Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Biology Paper 2 Question And Answer Compatibility with Devices
 - Biology Paper 2 Question And Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biology Paper 2 Question And Answer
 - Highlighting and Note-Taking Biology Paper 2 Question And Answer
 - Interactive Elements Biology Paper 2 Question And Answer
- 8. Staying Engaged with Biology Paper 2 Question And Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology Paper 2 Question And Answer
- 9. Balancing eBooks and Physical Books Biology Paper 2 Question And Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology Paper 2 Question And Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Biology Paper 2 Question And Answer
 - Setting Reading Goals Biology Paper 2 Question And Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Biology Paper 2 Question And Answer
 - Fact-Checking eBook Content of Biology Paper 2 Question And Answer
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web Feb 25 2022 detailed experimental light brown nutmeg powder 4 g was mixed with 25 ml of clear and colorless diethyl ether solvent and the brown cloudy mixture was filtered by gravity filtration with fluted fast flow filter paper in a powder funnel into a 250 ml round bottom flask rbf leaving light brown residue and beige filtrate diethyl ether was

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web trimyristin in the major triglyceride found in nutmeg representing 20-25% of the dried weight of ground nutmeg to extract the trimyristin we performed a solid liquid extraction using the organic solvent methylene chloride CH_2Cl_2

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web isolation of trimyristin from nutmeg purpose the purpose of this experiment is to isolate myristic acid from ground nutmeg seeds nutmeg seeds are extracted using tBME to yield crude trimyristin then recrystallized to yield pure trimyristin which will be hydrolyzed and undergo esterification

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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyristin is a triglyceride what are triglycerides prelab question 2 how do microwaves heat microwave procedure add approximately 2.5 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web in this week's lab experiment the purpose of this lab is to use new techniques to isolate the chemical trimyristin from the common spice nutmeg this lab requires five techniques: simple distillation, vacuum filtration, melting point determination, reflux, and extraction

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web hatice kocataş bursa uludağ university faculty of arts and sciences department of chemistry 16059 bursa turkey abstract in the present work a surfactant assisted convenient extraction method was developed for the isolation of trimyristin from nutmeg

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web natural products chemistry the isolation of trimyristin from nutmeg over 40% of the medicinal chemicals used throughout the developed world today were originally isolated from natural sources these sources include flowering plants, fungi, bacteria, and to a lesser extent animals especially marine animals

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web nov 1 2021 the skill of isolating trimyristin will be useful for students to obtain natural whitening raw materials derived from nutmeg various studies on the isolation can be used in natural products

experiment 6 lab report experiment 6 isolation of trimyristin - Jun 16 2023

web oct 18 2019 extraction is made a lot easier when extracting trimyristin from nutmeg because of its large quantity in nutmeg the nutmeg is moved from a solid to a liquid phase which is then used for extracting and isolating the trimyristin
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web trimyristin compound from nutmeg seed trimyristin compounds contained in nutmeg seed is very important to be isolated because it has various benefits trimyristin along with myristic acid myristicin and elemicin have antioxidant anticonvulsant analgesic anti inflammatory antidiabetic antibacterial and anti fungal activity

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web aug 4 2008 the purpose of this experiment was extract isolate and purify the natural product trimyristin from the spice nutmeg the trimyristin was extracted by using the solvent diethyl ether and

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

lab 4 extraction of trimyristin from nutmeg university of toronto - Sep 07 2022

web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

lab experiment 3 trimyristin extraction from nutmeg docsity - Feb 12 2023

web may 1 2011 experiment 3 1 february 2010 grade a small scale extraction of trimyristin from nutmeg introduction the purpose of this experiment is to collect trimyristin from an amount of nutmeg provided trimyristin is a type of triglyceride containing the fatty acid myristic acid also known as tetradecanoic acid

nutmeg lab final report isolation and purification of trimyristin - May 03 2022

web final report isolation and purification of trimyristin from nutmeg 2 purpose the purpose of this experiment is to isolate a chemical in nutmeg called trimyristin through extraction and to perform ir spectroscopy on this sample to determine its functional groups chemical structures signatures