

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

Chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a homogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution.

1. Define the term solvent and provide an example.

2. Define solubility, and how is it measured?

3. Distinguish between concentrated and dilute.

4. List the properties of bases and give an example.

5. Explain the significance of the pH scale.

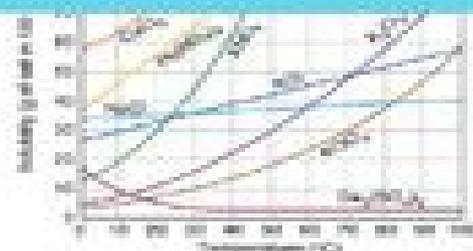
6. How does the concentration of an acid affect its pH?

7. Explain the process of neutralization.

8. Write the general equation for an acid-base reaction.

9. Explain the concept of a buffer solution.

10. Describe the properties of a strong acid and base.



11. Describe how you would use the graph to determine the amount of Na_2CO_3 that can be dissolved in 100 grams of water at 40°C.

12. Which salt has the highest solubility at 40°C?



ANSWER KEY INCLUDED

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T. W. Graham Solomons



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