

## **CHM 101: General Chemistry I (2,Units)**

### **Learning Outcomes**

At the end of this course, the students should be able to:

1. define atom, molecules and chemical reactions;
2. discuss the Modern electronic theory of atoms;
3. write electronic configurations of elements on the periodic table;
4. justify the trends of atomic radii, ionization energies, electronegativity of the elements based on their position in the periodic table;
5. identify and balance oxidation – reduction equation and solve redox titration problems;
6. illustrate shapes of simple molecules and hybridized orbitals;
7. identify the characteristics of acids, bases and salts, and solve problems based on their quantitative relationship;
8. apply the principles of equilibrium to aqueous systems using LeChatelier's principle to predict the effect of concentration, pressure and temperature changes on equilibrium mixtures;
9. analyse and perform calculations with the thermodynamic functions, enthalpy, entropy and free energy; and
10. determine rates of reactions and its dependence on concentration, time and temperature.

### **Course Contents**

Atoms, molecules, elements and compounds and chemical reactions. Modern electronic theory of atoms. Electronic configuration, periodicity and building up of the periodic table. Hybridization and shapes of simple molecules. Valence Forces and structure of solids. Chemical equations and stoichiometry, chemical bonding and intermolecular forces, kinetic theory of matter. Elementary thermochemistry, rates of reaction, equilibrium and thermodynamics. Acids, bases and salts. Properties of gases. Redox reactions and introduction to electrochemistry. Radioactivity.

# Chm 101 Noun Course Material

**JL Elias**



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**Resources in Education** ,1987 *The Distance Learning Playbook for College and University Instruction* Douglas Fisher,Nancy Frey,John Almarode,John Hattie,2020-08-20 First let s commend ourselves how in the midst of a pandemic we faculty stepped up at record speed to teach in such a foreign learning environment Try we did adapt we did and learn we did But to be clear and we already recognize this this past spring was less about distance learning and more about crisis teaching This time around we have the opportunity to be much more purposeful and intentional and that s where The Distance Learning Playbook for College and University Instruction will prove absolutely indispensable Much more than a collection of cool tools and apps The Distance Learning Playbook for College and University Instruction mobilizes decades of Visible Learning research to reveal those evidence based strategies that work best in an online environment Supplemented by video footage and opportunities to self assess and reflect the book addresses every dynamic that must be in place for students to learn even at a distance Faculty student relationships from a distance Teacher credibility from a distance Teacher clarity from a distance Engaging tasks from a distance Planning learning experiences from a distance Feedback assessment and grading from a distance Keeping the focus on learning from a distance or otherwise What does our post COVID future hold We suspect Fisher Frey Almarode and Hattie write it will include increased amounts of distance learning In the meantime let s seize on what we have learned to improve post secondary education in any format whether face to face or from a distance We are all still active faculty members committed to teaching scholarship and service The unexpected transition to remote learning doesn t mean we no longer know how to teach We can still impact the lives of our students and know that we made a difference The Distance Learning Playbook for College and University Instruction will show you how Douglas Fisher Nancy Frey John Almarode and John Hattie

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