

## Intermediate Cisco Networking interview questions and answers to ask mid-tier network engineers

- What is the purpose of HSRP, and how does it work to provide redundancy in a network?
- Can you explain the differences between different types of network topologies, such as star, ring, and mesh?
- How would you configure quality of service (QoS) on a Cisco device, and why is it important?
- What is the function of the Access Control List (ACL) in network security, and how would you implement one?
- Describe the process of configuring and managing a site-to-site VPN on a Cisco router.

# Cisco Engineer Interview Questions

**Lauren Gardner**



## **Cisco Engineer Interview Questions:**

## Decoding **Cisco Engineer Interview Questions**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Cisco Engineer Interview Questions**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of  $1s^2 2s^1$

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the

atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium  $1s^2 2s^2$  is an s electron whereas the electron removed during the ionization of boron  $1s^2 2s^2 2p^1$  is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he  $2$  helium  $1s^2$  ne  $10$  neon  $1s^2 2s^2 2p^6$

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$  or simply put as  $4s^1$  q2 what is the total number of electrons that can be filled in s p and d subshells

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