

# Oxidation-Reduction Titrations I: Determination of Oxalate

## B. Standardization of $\text{KMnO}_4$

	Trial 1	Trial 2	Trial 3
Mass of $\text{Na}_2\text{C}_2\text{O}_4$			
Weighing bottle initial mass	<u>2.4331</u>	<u>2.4331</u>	<u>2.4331</u>
Weighing bottle final mass	<u>2.6334</u>	<u>2.6358</u>	<u>2.6340</u>
Mass of $\text{Na}_2\text{C}_2\text{O}_4$	<u>0.2003</u>	<u>0.2027</u>	<u>0.2019</u>
Titration volume of $\text{KMnO}_4$			
Final reading	<u>26.4 mL</u>	<u>25.8 mL</u>	<u>16.5 mL</u>
Initial reading	<u>0.00 mL</u>	<u>0.00 mL</u>	<u>0.00 mL</u>
Volume of $\text{KMnO}_4$	<u>26.4 mL</u>	<u>25.8 mL</u>	<u>16.5 mL</u>

### Calculations

mMoles of $\text{Na}_2\text{C}_2\text{O}_4$	_____	_____	_____
Volume of $\text{KMnO}_4$	_____	_____	_____

mMoles of $\text{KMnO}_4$	_____	_____	_____
Molarity of $\text{KMnO}_4$	_____	_____	_____

Average molarity \_\_\_\_\_

Standard deviation (show calculations) \_\_\_\_\_

# Chemfax Lab Answers Oxidation Reduction Titrations

**David Kirk**



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