

CHEMISTRY STUDY GUIDE CHAPTER 15

Draw a water molecule with partial positives and negatives and explain why it is polar

Draw two water molecules with hydrogen bonds between them

Explain the net results that hydrogen bonding has on the properties of water in terms of surface tension, stickiness, shape, temperature and what happens to bonds when changing from liquid to gas

Explain what happens to the density of water from 100C to 4C to 0C

Explain the crystalline pattern of ice compared to water in terms of its density

Understand the difference between solutes and solvents and aqueous solutions

Polar solvents dissolve which two types of molecules

What it means to be soluble and insoluble

Know effects of heating/cooling or increasing/decreasing concentration has on solubility

Know the concept of "like dissolves like" to determine which substances will dissolve in others

Chemistry Solvation And Solubility Study Guide

**John Krenos, Joseph Potenza, Carl
Hoeger**



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