

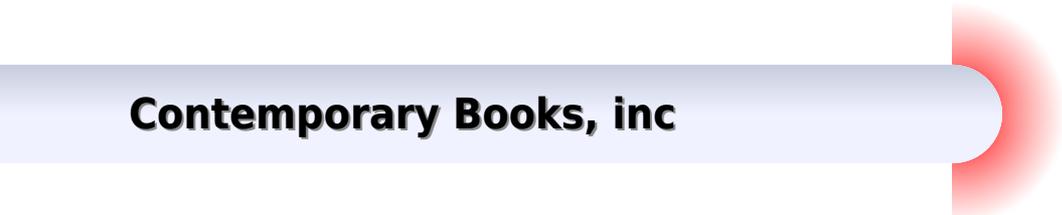
Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
a) **White**
b) Cream
c) Yellow
d) Green
2. What colour do sodium ions go in a flame test?
a) Red
b) **Yellow**
c) Brick-red
d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
a) Silver chloride and nitric acid
b) Barium chloride and hydrochloric acid
c) **Silver nitrate and nitric acid**
d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
a) Lit splint goes out with a squeaky pop
b) Damp blue litmus paper goes red
c) Damp red litmus paper goes blue
d) **Glowing splint relights**
5. What colour precipitate do iodide ions form?
a) White
b) Cream
c) **Yellow**
d) Green
6. Which ion produces a brick-red colour in a flame test?
a) Sodium
b) **Calcium**
c) Copper
d) Lithium
7. What is the test for carbon dioxide gas?
a) **Lime water turns cloudy**
b) Damp blue litmus turns red
c) A glowing splint relights
d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
a) White
b) Blue
c) Green
d) **Red/brown**
9. What is the test for chlorine gas?
a) **Damp blue litmus paper turns red and is then bleached white**
b) Red litmus \square turns blue
c) A glowing splint relights
d) Lime water will turn cloudy
10. Copper (II) ions form what colour precipitate test?
a) **Blue**
b) Green
c) Red/brown
d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
a) Silver chloride and nitric acid
b) Barium chloride and sulphuric acid
c) Silver nitrate and nitric acid
d) **Barium chloride and hydrochloric acid**
12. What colour do potassium ions go in a flame test?
a) **Lilac**
b) Red
c) Yellow
d) Brick-red
13. What is the test for hydrogen gas?
a) **Lit splint goes out with a squeaky pop**
b) Glowing splint relights
c) Damp blue litmus paper turns red and then is bleached white
d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
a) Sodium
b) Calcium
c) Copper
d) **Lithium**
15. What colour precipitate do bromide ions form?
a) White
b) **Cream**
c) Yellow
d) Green
16. What needs to be added to test for the presence of carbonate ions?
a) Silver nitrate
b) Universal indicator
c) Litmus paper
d) **Dilute acid**
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
a) Oxygen
b) **Ammonia**
c) Carbon dioxide
d) Hydrogen
18. What colour precipitate does barium sulphate form?
a) **White**
b) Green
c) Yellow
d) Cream
19. What is needed to carry out a flame test?
a) Splint
b) Glass rod
c) Spatula
d) **Nichrome/platinum wire**

Your score out of 19=
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Table of Contents Chemistry Unit 2 Test Answers

1. Understanding the eBook Chemistry Unit 2 Test Answers
 - The Rise of Digital Reading Chemistry Unit 2 Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit 2 Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Unit 2 Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit 2 Test Answers
 - Personalized Recommendations
 - Chemistry Unit 2 Test Answers User Reviews and Ratings
 - Chemistry Unit 2 Test Answers and Bestseller Lists
5. Accessing Chemistry Unit 2 Test Answers Free and Paid eBooks
 - Chemistry Unit 2 Test Answers Public Domain eBooks
 - Chemistry Unit 2 Test Answers eBook Subscription Services

- Chemistry Unit 2 Test Answers Budget-Friendly Options
- 6. Navigating Chemistry Unit 2 Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Unit 2 Test Answers Compatibility with Devices
 - Chemistry Unit 2 Test Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Unit 2 Test Answers
 - Highlighting and Note-Taking Chemistry Unit 2 Test Answers
 - Interactive Elements Chemistry Unit 2 Test Answers
- 8. Staying Engaged with Chemistry Unit 2 Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Unit 2 Test Answers
- 9. Balancing eBooks and Physical Books Chemistry Unit 2 Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Unit 2 Test Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Unit 2 Test Answers
 - Setting Reading Goals Chemistry Unit 2 Test Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Unit 2 Test Answers
 - Fact-Checking eBook Content of Chemistry Unit 2 Test Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

-
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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